=> fil reg

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STRUCTURE FILE UPDATES: 18 MAR 2009 HIGHEST RN 1123341-06-9
DICTIONARY FILE UPDATES: 18 MAR 2009 HIGHEST RN 1123341-06-9

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http://www.cas.org/support/stngen/stndoc/properties.html

=> d que stat 139 L3 STR

0~~C 1H2=2~Ak

VAR G1=X/7
NODE ATTRIBUTES:
DEFAULT MLEVEL IS ATOM
GGCAT IS SAT AT 3
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 6

STEREO ATTRIBUTES: NONE

្=្

NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 2

STEREO ATTRIBUTES: NONE

L5 SCR 1992 OR 2021 OR 2016 OR 2026 OR 2043 OR 1838

L6 29603 SEA FILE=REGISTRY SSS FUL L3 NOT L4 NOT L5

L37 STR

VAR G1=X/7/9

REP G2=(1-2) 15-2 16-14

REP G3=(0-2) 26-23 28-25

VAR G4=1/21

NODE ATTRIBUTES: CONNECT IS E1 RC AT 8

CONNECT IS E1 RC AT 12 CONNECT IS E1 RC AT 17 CONNECT IS E1 RC AT 18 CONNECT IS E1 RC AT 19 CONNECT IS E1 RC AT 27

CONNECT IS E1 RC AT 29 CONNECT IS E1 RC AT 30

DEFAULT MLEVEL IS ATOM GGCAT IS SAT AT 8 GGCAT IS SAT AT 12 GGCAT IS SAT AT 17 GGCAT IS SAT AT 18

GGCAT IS SAT AT 19 GGCAT IS SAT AT 27 GGCAT IS SAT AT 29

GGCAT IS SAT AT 30 DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 28

STEREO ATTRIBUTES: NONE

237 SEA FILE-REGISTRY SUB-L6 SSS FUL L37

100.0% PROCESSED 29603 ITERATIONS 237 ANSWERS SEARCH TIME: 00.00.39

=> d que stat 127 L20 STR March 20, 2009 10/559,779 3

VAR G1=1/7
VAR G2=X/15
NODE ATTRIBUTES:
DEFAULT MLEVEL IS ATOM
GGCAT IS SAT AT 13
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RSPEC I

NUMBER OF NODES IS 16

STEREO ATTRIBUTES: NONE
L22 SCR 1992 OR 2021 OR 2016 OR 2026
L24 41607 SEA FILE=REGISTRY SSS FUL L20 NOT L22
L25 STR

Page 1-B
VAR C1=21/23
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REP G3=(0-3) 27-6 28-8
REP G4=(0-3) 31-12 33-18
VAR G5=5/16
VAR G5=5/16
CONNECT IS E1 RC AT 9
CONNECT IS E1 RC AT 19
CONNECT IS E1 RC AT 22
CONNECT IS E1 RC AT 22
CONNECT IS E1 RC AT 26
CONNECT IS E1 RC AT 27
CONNECT IS E1 RC AT 27
CONNECT IS E1 RC AT 32
CONNECT IS E1 RC AT 32
CONNECT IS E1 RC AT 33

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CONNECT IS E1 RC AT 34
DEFAULT MLEVEL IS ATOM
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GGCAT IS SAT AT 22
GGCAT IS SAT AT 26
GGCAT IS SAT AT 29
GGCAT IS SAT AT 30
GGCAT IS SAT AT 32
GGCAT IS SAT AT 34
DEFAULT ECLEVEL IS LIMITED
GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS 35
STEREO ATTRIBUTES: NONE
L27
          362 SEA FILE=REGISTRY SUB=L24 SSS FUL L25
100.0% PROCESSED 41607 ITERATIONS
                                                           362 ANSWERS
SEARCH TIME: 00.00.01
=> d his
     (FILE 'HOME' ENTERED AT 10:59:30 ON 20 MAR 2009)
    FILE 'HCAPLUS' ENTERED AT 10:59:39 ON 20 MAR 2009
            1 S 2004:1154740/AN
    FILE 'REGISTRY' ENTERED AT 11:00:00 ON 20 MAR 2009
              ACT HUH779AU/A
L2
            14 SEA FILE=REGISTRY ABB=ON PLU=ON (10294-34-5/BI OR 110-8
               ACT HUH779/A
L3
               STR
L4
               STR
L5
               SCR 1992 OR 2021 OR 2016 OR 2026 OR 2043 OR 1838
L6
         29603 SEA FILE=REGISTRY SSS FUL L3 NOT L4 NOT L5
    FILE 'HCAPLUS' ENTERED AT 11:11:51 ON 20 MAR 2009
L7
           876 S L6(L)CAT/RL
L8
               QUE POLYISOBUT?
L9
            18 S L7 AND L8
T-10
            17 S L9 NOT L1
    FILE 'LREGISTRY' ENTERED AT 11:13:46 ON 20 MAR 2009
L11
               STR L3
    FILE 'REGISTRY' ENTERED AT 11:21:57 ON 20 MAR 2009
L12
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L13
               STR 1.11
L14
             6 S L13 SSS SAM SUB=L6
L15
           109 S L13 SSS FUL SUB=L6
               SAV L15 HUH779S1/A
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FILE 'HCAPLUS' ENTERED AT 11:25:36 ON 20 MAR 2009

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L17
            3 S L7 AND L16
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1.18
              STR
    FILE 'REGISTRY' ENTERED AT 11:37:24 ON 20 MAR 2009
L19
           50 S L18
L20
              STR L18
L21
            50 S L20
L22
              SCR 1992 OR 2021 OR 2016 OR 2026
T.23
            50 S L20 NOT L22
L24
        41607 S L20 NOT L22 FUL
               SAV TEMP L24 HUH779A/A
    FILE 'LREGISTRY' ENTERED AT 11:43:24 ON 20 MAR 2009
1.25
              STR
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L27
           362 S L25 SSS FUL SUB=L24
              SAV L27 HUH779S2/A
   FILE 'HCAPLUS' ENTERED AT 11:53:31 ON 20 MAR 2009
         2011 S L27
              QUE CAT# OR CATALYST?
L29
L30
           46 S L16 AND L29
L31
             OUE LEWIS(N) ACID?
L32
            5 S L30 AND L31
L33
            8 S L17 OR L32
L34
          135 S L28 AND L29
            5 S L34 AND L31
L35
L36
            0 S L27(L)CAT/RL
   FILE 'REGISTRY' ENTERED AT 11:57:19 ON 20 MAR 2009
1.37
             STR L13
L38
           11 S L37 SSS SAM SUB=L6
          237 S L37 SSS FUL SUB=L6
L39
              SAV L39 HUH779S3/A
    FILE 'HCAPLUS' ENTERED AT 11:59:15 ON 20 MAR 2009
L40
            2 S L39(L)CAT/RL
L41
          391 S L39
L42
          91 S L41 AND L29
L43
           14 S L42 AND L31
L44
           17 S L33 OR L43
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FILE COVERS 1907 - 20 Mar 2009 VOL 150 ISS 13 FILE LAST UPDATED: 19 Mar 2009 (20090319/ED)

HCAplus now includes complete International Patent Classification (IPC) reclassification data for the third quarter of 2008.

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http://www.cas.org/legal/infopolicy.html

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> d ibib abs hitstr hitind 144 1-17

L44 ANSWER 1 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2008:1525040 HCAPLUS Full-text
DOCUMENT NUMBER: 150:213601

TITLE: Lewis acid catalyst

111111.

free electrophilic alkylation of silicon-capped
π donors in 1,1,1,3,3,3-hexafluoro-2-propanol
AUTHOR(S): Ratnikov, Maxim Ο.; Tumanov, Vasily V.; Smit,

William A.

CORPORATE SOURCE: N. D. Zelinsky Institute of Organic Chemistry, Moscow, 119991, Russia

MOSCOW, 119991, RUSSIA

SOURCE: Angewandte Chemie, International Edition (2008), 47(50), 9739-9742

CODEN: ACIEF5; ISSN: 1433-7851

PUBLISHER: Wiley-VCH Verlag GmbH & Co. KGaA

DOCUMENT TYPE: Journal LANGUAGE: English

AB A diverse range of electrophilic addition reactions, conventionally carried out using Lewis acid catalysts, have been performed in electrophilic media without acidic reagents. 1,1,1,3,3,3-Hexafluoro-2-propanol (HFIP) acts as an efficient hydrogen-bond donor and highly polar solvent for a wide range of

substrates in reactions with Si-capped π donors.

IT 68105-14-6P 152684-80-5P

RL: SPN (Synthetic preparation); PREP (Preparation) (1,1,1,3,3,3-hexafluoro-2-propanol-mediated electrophilic alkylation of silicon-capped π donors by acetals and carbonyl compds. in the absence of Lewis acid

catalysts)
RN 68105-14-6 HCAPLUS

CN 1-Hexene, 4-methoxy-2,5-dimethyl- (CA INDEX NAME)

RN 152684-80-5 HCAPLUS

CN 1-Hexene, 4-methoxy-5-methyl- (CA INDEX NAME)

```
H 2 C CH CH 2 CH Pr - 1
    21-2 (General Organic Chemistry)
ТТ
    Acetals
     Aldehydes, reactions
     Carbonyl compounds (organic), reactions
     Ketones, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (1,1,1,3,3,3-hexafluoro-2-propanol-mediated electrophilic
        alkylation of silicon-capped \pi donors by acetals and carbonyl
        compds. in the absence of Lewis acid
        catalysts)
    Alkylation
        (electrophilic; 1,1,1,3,3,3-hexafluoro-2-propanol-mediated
        electrophilic alkylation of silicon-capped \pi donors by acetals
        and carbonyl compds. in the absence of Lewis
        acid catalysts)
     Silanes
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (unsatd.; 1,1,1,3,3,3-hexafluoro-2-propanol-mediated
        electrophilic alkylation of silicon-capped \pi donors by acetals
        and carbonyl compds. in the absence of Lewis
        acid catalysts)
   Electron donors
        (π-donors; 1,1,1,3,3,3-hexafluoro-2-propanol-mediated
        electrophilic alkylation of silicon-capped \pi donors by acetals
        and carbonyl compds. in the absence of Lewis
        acid catalysts)
     111874-55-6P
                   165059-61-0P
     RL: BYP (Byproduct); PREP (Preparation)
        (1,1,1,3,3,3-hexafluoro-2-propanol-mediated electrophilic
        alkylation of silicon-capped \pi donors by acetals and carbonyl
       compds. in the absence of Lewis acid
        catalvsts)
     920-66-1, 1,1,1,3,3,3-Hexafluoro-2-propanol
     RL: NUU (Other use, unclassified); RGT (Reagent); RACT (Reactant or
     reagent); USES (Uses)
        (1,1,1,3,3,3-hexafluoro-2-propanol-mediated electrophilic
        alkylation of silicon-capped \pi donors by acetals and carbonyl
        compds. in the absence of Lewis acid
        catalvsts)
    78-84-2, Isobutanal
                         78-94-4, Methyl vinyl ketone, reactions
     96-33-3, Methyl acrylate 100-42-5, Styrene, reactions
     Anisic aldehyde, reactions 762-72-1, Allyltrimethylsilane
    930-30-3, 2-Cyclopentenone 931-94-2, Cyclopentanone dimethyl
     acetal
             933-01-7, p-Chlorophenylsulfenyl chloride 1125-88-8,
     Benzaldehyde dimethyl acetal 1833-53-0, 2-Trimethylsilyloxypropene
     6651-36-1, 1-Trimethylsilyloxycyclohexene
                                               13735-81-4.
     α-Trimethylsilyloxystyrene 17510-46-2,
    3,3-Dimethy1-2-trimethylsilyloxy-1-butene
                                               18292-38-1.
    Methallvltrimethylsilane 19980-43-9,
    1-(Trimethylsilyloxy)cyclopentene 31469-15-5
     RL: RCT (Reactant); RACT (Reactant or reagent)
```

(1,1,1,3,3,3,3-hexafluoro-2-propanol-mediated electrophilic alkylation of silicon-capped π donors by acetals and carbonyl compds. in the absence of Lewis acid caralysts)

IT 1489-27-6P, 2-(3-Oxobuty1)cyclopentanone 1501-04-8P, Methy1 5-0x0-5-phenylpentanoate 13051-32-6P, Dimethy1

2,2-dimethylpentanedioate 22039-97-0P 68105-14-6P 74209-73-7P 74209-74-8P 75359-63-6P 86289-92-1P, Methyl

73239-05-06 86289-92-1F, M 2,2-dimethyl-5-oxohexanoate 89597-27-3P 111874-54-5P 134111-20-9P 152684-80-5P 761411-20-5P 1114963-11-9

1114963-15-3P 1114963-16-4P 1114963-18-6P

RL: SPN (Synthetic preparation); PREP (Preparation)

(1,1,1,3,3,3)-hexafluoro-2-propanol-mediated electrophilic alkylation of silicon-capped π donors by acetals and carbonyl comods. in the absence of Lewis acid

catalysts)

REFERENCE COUNT: 33 THERE ARE 33 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

DOCUMENT NUMBER: 148:309625

TITLE: Arborescent polymers based on at least one inimer and isoolefin that have been

end-functionalized with a polymer or copolymer having a low glass transition temperature

INVENTOR(S): Kaszas, Gabor; Puskas, Judit; Kulbaba, Kevin; Nelson, Robert Lloyd

PATENT ASSIGNEE(S): The University of Akron, USA; Lanxess, Inc.

SOURCE: PCT Int. Appl., 76pp.
CODEN: PIXXD2

DOCUMENT TYPE: Patent
LANGUAGE: English
FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT I	.00		KIN	D	DATE			APPL	ICAT:	I NOI	NO.		D	ATE
		-												
WO 2008			A2		2008	0306		WO 2	007-	US19:	280			
													3	00708 1
WO 2008	027589		A3		2008	0424								
W:	AE, AG													
	CA, CH,	CN,	co,	CR,	CU,	CZ,	DE,	DK,	DM,	DO,	DZ,	EC,	EE,	EG,
	ES, FI,	GB,	GD,	GE,	GH,	GM,	GT,	HN,	HR,	HU,	ID,	IL,	IN,	IS,
	JP, KE,	KG,	KM,	KN,	KP,	KR,	KZ,	LA,	LC,	LK,	LR,	LS,	LT,	LU,
	LY, MA,	MD,	ME,	MG,	MK,	MN,	MW,	MX,	MY,	MZ,	NA,	NG,	NI,	NO,
	NZ, OM,	PG,	PH,	PL,	PT,	RO,	RS,	RU,	SC,	SD,	SE,	SG,	SK,	SL,
	SM, SV	SY,	TJ,	TM,	TN,	TR,	TT,	TZ,	UA,	UG,	US,	UZ,	VC,	VN,
	ZA, ZM	ZW												
RW:	AT, BE	BG,	CH,	CY,	CZ,	DE,	DK,	EE,	ES,	FI,	FR,	GB,	GR,	HU,
	IE, IS,	IT,	LT,	LU,	LV,	MC,	MT,	NL,	PL,	PT,	RO,	SE,	SI,	SK,
	TR, BF,	BJ,	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,	NE,	SN,
	TD, TG,	BW,	GH,	GM,	KE,	LS,	MW,	MZ,	NA,	SD,	SL,	SZ,	TZ,	UG,
	ZM, ZW,	AM,	AZ,	BY,	KG,	KZ,	MD,	RU,	TJ,	TM,	AP,	EA,	EP,	OA
PRIORITY APP	LN. INFO).:						US 2	006-	8417	57P	1	₽	

AB Disclosed is an arborescent polymer comprising: an elastomeric polymer portion having two or more branching points, the arborescent elastomeric polymer block having a low glass transition temperature (Tq), and one or more endfunctionalized portions, wherein one or more end-functionalized portions terminate at least one or two or more branches of the arborescent elastomeric polymer portion of the end-functionalized arborescent polymer. In one embodiment, the present invention relates to arborescent polymers formed from at least one inimer and at least one isoolefin that have been endfunctionalized with a polymer or copolymer having a low glass transition temperature (Tq), and to a process for making such arborescent polymers. In another embodiment, the present invention relates to arborescent polymers formed from at least one inimer and at least one isoolefin that have been endfunctionalized with less than about 5 weight% end blocks derived from a polymer or copolymer having a high glass transition temperature (Tg), and to a process for making such arborescent polymers. In an example of the invention 0.35 g of p-methoxycumylstyrene, 900 cm3 hexane, 600 cm3 Me chloride, 2 cm3 2,6-di-tert-butylpyridine and 240 cm3 isobutylene were added to a reactor, polymerization was started at -95° by addition of 6 cm3 of TiCl4, after 2 h of polymerization a mixture of 236 cm3 isoprene and 150 cm3 Me chloride, 150 cm3 hexane and a small amount of 2,6-di-tert-butylpyridine were added, and the polymerization was continued for another 30 min, before terminating it by adding methanol and NaOH to the mixture, after Me chloride evaporation, washing and isolation of the polymer, the dried weight of the polymer was 164.7 q. Polymers obtained in a similar way were mixed with carbon black, and optionally with a vulcanization agent, and tested as such and with the filler in comparison with regular butyl rubber.

IT 69078-25-7DP, arborescent polymers/synthetic rubbers 477350-05-3DP, arborescent polymers/synthetic rubbers 477350-08-6DP, arborescent polymers/synthetic rubbers 477350-12-2DP, arborescent polymers/synthetic rubbers RL: IMF (Industrial manufacture); PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses) (arborescent polymers based on branched functionalized styrene

inimer and isoolefins, like isobutylene) RN 69078-25-7 HCAPLUS 1-Hexene, 5-chloro-3,3,5-trimethyl- (CA INDEX NAME)

CN

RN 477350-05-3 HCAPLUS CN 1-Heptene, 6-chloro-4, 4, 6-trimethyl- (CA INDEX NAME)

CN 1-Nonene, 8-chloro-4, 4, 6, 6, 8-pentamethyl- (CA INDEX NAME)

RN 477350-12-2 HCAPLUS

CN 1-Octene, 7-chloro-3,3,5,5,7-pentamethyl- (CA INDEX NAME)

CC 39-4 (Synthetic Elastomers and Natural Rubber)

IT Polymerization catalysts

(Lewis acid, halides; arborescent polymers based on branched functionalized styrene inimer and isoolefins,

like isobutylene) 78-79-5DP, arborescent polymers/synthetic rubbers 100-42-5DP, arborescent polymers/synthetic rubbers 100-42-5DP, derivs., arborescent polymers/synthetic rubbers 106-99-0DP, 1,3-Butadiene, arborescent polymers/synthetic rubbers 592-46-1DP, 2,4-Hexadiene, arborescent polymers/synthetic rubbers 763-29-1DP, arborescent polymers/synthetic rubbers 763-29-1DP, arborescent polymers/synthetic rubbers 763-30-4DP, arborescent polymers/synthetic rubbers 764-13-6DP, arborescent polymers/synthetic rubbers 1118-58-7DP, arborescent polymers/synthetic rubbers 2622-21-1DP, arborescent polymers/synthetic rubbers 3049-88-5DP, arborescent polymers/synthetic rubbers 4049-81-4DP, arborescent polymers/synthetic rubbers 4549-74-0DP, arborescent polymers/synthetic rubbers 5732-01-4DP, 1,3-Butadiene, 2-neopentyl-, arborescent polymers/synthetic rubbers 13643-06-6DP. arborescent polymers/synthetic rubbers 25924-78-1DP, Pipervline, arborescent polymers/synthetic rubbers 26519-91-5DP, Methylcyclopentadiene, arborescent polymers/synthetic rubbers 29797-09-9DP, Cyclohexadiene, arborescent polymers/synthetic rubbers 57908-07-3DP, arborescent polymers/synthetic rubbers 69078-25-7DP, arborescent polymers/synthetic rubbers 140220-46-8DP, arborescent polymers/synthetic rubbers 140220-95-7DP, arborescent polymers/synthetic rubbers 173932-32-6DP, arborescent polymers/synthetic rubbers 477350-02-0DP, arborescent polymers/synthetic rubbers 477350-03-1DP, arborescent polymers/synthetic rubbers 477350-04-2DP, arborescent polymers/synthetic rubbers 477350-05-3DP, arborescent polymers/synthetic rubbers 477350-06-4DP, arborescent polymers/synthetic rubbers 477350-07-5DP, arborescent polymers/synthetic rubbers 477350-08-6DP, arborescent polymers/synthetic rubbers 477350-09-7DP, arborescent polymers/synthetic rubbers

477350-10-0DP, arborescent polymers/synthetic rubbers

477350-11-1DP, arborescent polymers/synthetic rubbers 477350-12-2DP, arborescent polymers/synthetic rubbers 477350-13-3DP, arborescent polymers/synthetic rubbers RL: IMF (Industrial manufacture); PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses)

(arborescent polymers based on branched functionalized styrene inimer and isoolefins, like isobutylene)

L44 ANSWER 3 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2007:1380460 HCAPLUS Full-text DOCUMENT NUMBER: 148:35160

TITLE:

Material and process for precisely controlled

polymeric coatings INVENTOR(S): Huang, Xueving

PATENT ASSIGNEE(S): Sepax Technologies, Inc., USA

SOURCE: U.S., 12pp. CODEN: USXXAM

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 7303821	B1	20071204	US 2004-887629	200407
PRIORITY APPLN. INFO.:			US 2003-489786P P	10 200307

AB A chemical coated surface comprises: (a) a support surface; (b) a coupling agent; (c) a chemical surface coating comprising at least one block copolymer prepared from addition monomers; wherein the coupling agent is a polymer or an oligomer comprising of: (i) at least two functional groups capable of binding onto the support surface; (ii) at least two functional groups capable of generating a living radical for initiating polymerization of addition monomers; and wherein the coupling agent has a hydrocarbon backbone. The surface typically comprises of block copolymers prepared from addition monomers selected from a group consisting of acrylates and its derivs., metharvlates and its derivs., styrene and its derivs., acrylamide, methacrylamide, dimethacrylamide, N-monosubstituted acrylamide, Nmonosubstituted methacrylamide, N,N-disubstituted acrylamide, N,Nbissubstituted methacrylamide, vinyl acetate, vinyl pyrrolidone, vinyl ether, acrylic acid and methacrylic acid. This invention also provides a chemical process for making a chemical coated surface for the electrophoretic separation application.

24

213453-08-8P

RL: CAT (Catalyst use); IMF (Industrial manufacture); PREP (Preparation); USES (Uses)

(initiator; material and process for precisely controlled polymeric coatings)

- RN 213453-08-8 HCAPLUS
- 2-Propenoic acid, 2-methyl-, 2-(2-bromo-2-methyl-1-oxopropoxy)ethyl CN ester (CA INDEX NAME)

TT 959129-43-2P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (material and process for precisely controlled polymeric

coatings) RN 959129-43-2 HCAPLUS

CN 4-Penten-1-ol, 2-bromo-2-methyl-, 1-propanoate (CA INDEX NAME)

INCL 428446000; 526081000; 526087000; 526091000; 526109000; 526118000; 526201000

CC 42-10 (Coatings, Inks, and Related Products)

Section cross-reference(s): 79, 80

IT 213453-08-8P 707471-11-2P

RL: CAT (Catalyst use); IMF (Industrial manufacture); PREP (Preparation); USES (Uses) (initiator; material and process for precisely controlled polymeric coatings)

IT 959129-43-2P

RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (material and process for precisely controlled polymeric

coatings)

REFERENCE COUNT: 22 THERE ARE 22 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L44 ANSWER 4 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2004:479523 HCAPLUS Full-text

DOCUMENT NUMBER: 141:123433

TITLE: Allylation of acetals and ketals with allyltrimethylsilane catalyzed by the mixed Lewis acid system AlBr3/CuBr

AUTHOR(S): Jung, Michael E.; Maderna, Andreas

CORPORATE SOURCE: Department of Chemistry and Biochemistry, Los

Angeles, University of California, Los Angeles, CA, 90095-1569, USA

SOURCE: Tetrahedron Letters (2004), 45(27), 5301-5304

CODEN: TELEAY; ISSN: 0040-4039

PUBLISHER: Elsevier
DOCUMENT TYPE: Journal
LANGUAGE: English

OTHER SOURCE(S): CASREACT 141:123433

GT

March 20, 2009 10/559,779 13

- AB The synthesis of homoallyl ethers, e.g., I, from various acetals and a cyclic ketal with allyltrimethylsilane catalyzed by aluminum bromide in the presence of trimethylaluminum as a desiccant is described. The addition of a catalytic amount of copper(I) bromide accelerated the allylation reactions to afford the homoallyl ether derivs. in good to excellent yield.
- IT 105750-93-4P
 - RL: SPN (Synthetic preparation); PREP (Preparation)
 (preparation of homoallyl ethers via allylation of acetals and

cyclohexanone di-Me ketal with allyltrimethylsilane catalyzed by mixed Lewis acid system aluminum

tribromide-trimethylaluminum-copper bromide)

- RN 105750-93-4 HCAPLUS
- CN 1-Undecene, 4-methoxy- (CA INDEX NAME)

- CC 25-9 (Benzene, Its Derivatives, and Condensed Benzenoid Compounds)
 ST acetal allyltrimethylsilane allylation aluminum tribromide copper
- Si acetai aijyirimetnyisilane aijyiation aluminum tribromide copper bromide atalyst; homoallylic ether preph; copper bromide allylation catalyst; aluminum tribromide allylation catalyst
- IT Allylation

Allylation catalysts

(preparation of homoallyl ethers via allylation of acetals and cyclohexanone di-Me ketal with allyltrimethylsilane catalyzed by mixed Lewis acid system aluminum

tribromide-trimethylaluminum-copper bromide)

- T Acetals
 - RL: RCT (Reactant); RACT (Reactant or reagent)

(preparation of homoallyl ethers via allylation of acetals and cyclohexanone di-Me ketal with allyltrimethylsilane catalyzed by mixed Lewis acid system aluminum

tribromide-trimethylaluminum-copper bromide)

- IT Ethers, preparation
 - RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of homoallyl ethers via allylation of acetals and cyclohexanone di-Me ketal with allyltrimethylsilane catalyzed by mixed Lawis acid system aluminum

tribromide-trimethylaluminum-copper bromide)

IT 75-24-1, Trimethylaluminum 7727-15-3, Aluminum tribromide

7787-70-4, Copper(I) bromide

RL: CAT (Catalyst use); USES (Uses)

(preparation of homoallyl ethers via allylation of acetals and cyclohexanone di-Me ketal with allyltrimethylsilane catalyzed by mixed Lewis acid system aluminum

tribromide-trimethylaluminum-copper bromide) 101-48-4, Phenylacetaldehyde dimethyl acetal 762-72-1, Allvltrimethylsilane 933-40-4, Cyclohexanone dimethyl ketal 1125-88-8, Benzaldehyde dimethyl acetal 2186-92-7, 4-Methoxybenzaldehyde dimethyl acetal 7252-83-7, Bromoacetaldehyde dimethyl acetal 10022-28-3, Octanal dimethyl acetal 14618-78-1 24856-58-4, 4-Bromobenzaldehyde dimethyl acetal RL: RCT (Reactant); RACT (Reactant or reagent) (preparation of homoallyl ethers via allylation of acetals and cyclohexanone di-Me ketal with allyltrimethylsilane catalyzed by mixed Lewis acid system aluminum tribromide-trimethylaluminum-copper bromide)

22039-97-0P 60753-94-8P 77920-96-8P 105750-93-4P

111874-57-8P 114095-76-0P 124389-38-4P 724701-62-6P RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of homoallyl ethers via allylation of acetals and cyclohexanone di-Me ketal with allyltrimethylsilane catalyzed by mixed Lewis acid system aluminum

tribromide-trimethylaluminum-copper bromide)

REFERENCE COUNT: 16 THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L44 ANSWER 5 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2003:825944 HCAPLUS Full-text

DOCUMENT NUMBER: 140:78789

TITLE: Saucy-Marbet ketonization in a continuous

fixed-bed catalytic reactor AUTHOR(S): Frauchiger, Simon; Baiker, Alfons

CORPORATE SOURCE: Institute for Chemical and Bioengineering, Swiss

Federal Institute of Technology, ETH

Hoenggerberg, Zurich, CH-8093, Switz.

SOURCE: Applied Catalysis, A: General (2003), 253(1),

33 - 48

CODEN: ACAGE4; ISSN: 0926-860X

PUBLISHER: Elsevier Science B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 140:78789

Saucy-Marbet ketonizations of two unsatd. alcs., 3,7-dimethyl-oct-6-en-1-yn-3-ol (dehydrolinalool (DLL)) and 2-methyl-3-butyn-2-ol (MB) with 2-AB methoxypropene (MP), were studied in a continuous fixed-bed reactor containing a com. alkyl sulfonic acid polysiloxane catalyst, Deloxan ASP I/9, Degussa, a condensate of propyl-3-sulfonic acid-siloxane and SiO2. The solid acid catalyst showed an initial transient period of several hours until stable operation was reached. Similar products were observed as in corresponding homogeneous catalytic reactions indicating similar reaction pathways. However, the heterogeneous catalytic reactions showed lower selectivity to the desired products than the corresponding homogeneous reactions reported in the literature. This outcome is traced to the complex transformation the unsatd. ether MP undergoes upon interaction with the surface of the solid acid catalyst. Furthermore, both Bronsted and Lewis-acid sites were present on the catalyst surface. These factors are the main reasons for the less efficient use of reactants observed in the heterogeneous catalytic reactions. A maximum yield of 54% to the desired product could be achieved in the heterogeneous catalytic ketonization of 3,7-dimethyl-oct-6-en-1-yn-3-ol and 45% for the 2methyl-3-butyn-2-ol ketonization, resp. The studies indicate that the continuous ketonization in a fixed-bed reactor is feasible, but for tech. application, more selective acid catalysts are necessary to compete with the generally used homogeneous catalytic routes.

IT 191930-68-4P

RL: BYP (Byproduct); PREP (Preparation)
(conversion and selectivity of sulfonic acid polysiloxane/silica

catalyst in Saucy-Marbet ketonizations of unsatd. alcs. in continuous fixed-bed reactor)

RN 191930-68-4 HCAPLUS

CN 1-Pentene, 2,4-dimethoxy-4-methyl- (CA INDEX NAME)

CC 45-4 (Industrial Organic Chemicals, Leather, Fats, and Waxes) Section cross-reference(s): 67

ST unsatd alc ketonization methoxypropene heterogeneous catalytic route; dimethyloctenynol ketonization alkylsulfonic acid polysiloxane silica catalyst

IT Carbonylation catalysts

(Saucy-Marbet, heterogeneous; conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet ketonizations of unsatd. alcs. in continuous fixed-bed reactor)

Carbonylation

(Saucy-Marbet; conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet

ketonizations of unsatd. alcs. in continuous fixed-bed reactor) Lewis acidity

1 andward documents

(conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet ketonizations of unsatd. alcs. in continuous fixed-bed reactor)

IT Reactors

IT

(fixed-bed; conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet ketonizations of unsatd. alcs. in continuous fixed-bed reactor)

IT Polysiloxanes, uses

RL: CAT (Catalyst use); USES (Uses)

(propyl-3-sulfonic acid containing; conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in

Saucy-Marbet ketonizations of unsatd. alcs. in continuous fixed-bed reactor)

IT 77-76-9P, 2,2-Dimethoxypropane 141-10-6P 53654-62-9P 191930-68-4P

RL: BYP (Byproduct); PREP (Preparation)

(conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet ketonizations of unsatd. alcs.

in continuous fixed-bed reactor)

T 7631-86-9, Silica, uses 640735-92-8, Deloxan ASP-I 9 RL: CAT (Catalyst use); USES (Uses)

(conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet ketonizations of unsatd. alcs. in continuous fixed-bed reactor)

IT 115-19-5, 2-Methyl-3-butyn-2-ol 116-11-0 29171-20-8,

3,7-Dimethyl-oct-6-en-1-yn-3-ol

RL: EPR (Engineering process); PEP (Physical, engineering or chemical process); RCT (Reactant); PROC (Process); RACT (Reactant or reagent)

March 20, 2009 10/559,779 16

(conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet ketonizations of unsatd. alcs. in continuous fixed-bed reactor)

6,10-Dimethyl-4,5,9-undecatrien-2-one

RL: IMF (Industrial manufacture); PREP (Preparation)

(conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet ketonizations of unsatd. alcs.

in continuous fixed-bed reactor)

REFERENCE COUNT: 29 THERE ARE 29 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L44 ANSWER 6 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2002:927482 HCAPLUS Full-text

DOCUMENT NUMBER: 138:5461

TITLE: Arborescent polyolefinic thermoplastic elastomers and products therefrom

INVENTOR(S): Puskas, Judit E.; Paulo, Christophe; Antony,

Prince

PATENT ASSIGNEE(S): The University of Western Ontario, Can.

SOURCE: PCT Int. Appl., 46 pp.
CODEN: PIXXD2

DOCUMENT TYPE: Patent

LANGUAGE: English FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

					KIN		DATE		APPLICATION NO.						DATE	
WO 2002096967			2002096967 A1 2002120							WO 2002-CA792						00205
		CN, GE, LC, NO, TM, GH, CH, SE,	CO, GH, LK, NZ, TN, GM, CY,	CR, GM, LR, OM, TR, KE, DE, BF,	CU, HR, LS, PH, TT, LS, DK,	CZ, HU, LT, PL, TZ, MW, ES,	DE, ID, LU, PT, UA, MZ, FI,	DK, IL, LV, RO, UG, SD, FR,	DM, IN, MA, RU, UZ, SL, GB,	DZ IS MD SD VN SZ GR	BG, EC, JP, MG, SE, YU, TZ, IE, GN,	EE, KE, MK, SG, ZA, UG, IT,	ES, KG, MN, SI, ZM, ZM, LU,	FI, KP, MW, SK, ZW ZW, MC,	GB, KR, MX, SL, AI, NL,	GD, KZ, MZ, TJ, BE, PT,
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	2002									AU :	2002-	3022	69		2:	00205 9
	2002						2007 2003			us :	2002-	1567	67		2:	00205
	6747 1395						2004 2004			EP :	2002-	7297	29		2:	00205 9

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR JP 2004526857 20040902 200205 29 JP 4231777 B2 20090304 CN 1604923 A 20050406 CN 2002-811035 200205 29 PRIORITY APPLN. INFO.: US 2001-293508P 200105 29 WO 2002-CA792 200205

AB The present invention provides highly branched block copolymers comprising branched soft segments with a low Tg and hard segments with a high Tg or crystalline m.p. that exhibit thermoplastic elastomeric properties. More particularly the invention provides a highly branched block copolymer of a polyisoolefin and a polymonovinylidene arene characterized by having thermoplastic elastomeric properties wherein the aforesaid block copolymer comprises a highly branched polyisoolefin block and some of the branches of the aforesaid polyisoolefin block terminate in polymonovinylidene arene endblocks. Arborescent copolymers of isoolefins and monovinylidene arenes are prepared by (a) polymerizing an isoolefin (e.g., isobutylene) in the presence of at least one inimer (initiator-monomer) and a Lewis acid halide coinitiator, at -20 to - 100° to produce an arborescent elastomeric branched polymer, the inimer including at least one group for (co)polymerizing in a cationic polymerization of the isoolefin and at least one group for initiating cationic polymerization of the isoolefin; and thereafter (b) adding a compound having an effective electron pair donor for improving blocking efficiency and adding a monovinylidene arene suitable for the production of the polyvinylidene plastic blocks; and (c) terminating the polymerization reaction after a selected period of time by addition of an effective nucleophile compound which terminates the polymerization reaction, the selected period of time being sufficiently long enough to ensure production of at least individual units of the arborescent branched block copolymer.

69078-25-7, 5-Chloro-3,3,5-trimethyl-1-hexene 477350-05-3 477350-08-6 477350-12-2

RL: CAT (Catalyst use); USES (Uses)

(inimer; arborescent polyolefinic thermoplastic elastomers and

products therefrom)

RN 69078-25-7 HCAPLUS

CN 1-Hexene, 5-chloro-3,3,5-trimethyl- (CA INDEX NAME)

RN 477350-05-3 HCAPLUS

CN 1-Heptene, 6-chloro-4, 4, 6-trimethyl- (CA INDEX NAME)

March 20, 2009 10/559,779 18

RN 477350-08-6 HCAPLUS

CN 1-Nonene, 8-chloro-4,4,6,6,8-pentamethyl- (CA INDEX NAME)

RN 477350-12-2 HCAPLUS

CN 1-Octene, 7-chloro-3,3,5,5,7-pentamethvl- (CA INDEX NAME)

IC ICM C08F297-00

ICS C08G083-00

CC 39-4 (Synthetic Elastomers and Natural Rubber)

Section cross-reference(s): 63 3049-88-5 7005-39-2 57908-07-3 69078-25-7,

5-Chloro-3,3,5-trimethyl-1-hexene 140220-46-8 140220-95-7

173932-32-6 477350-02-0 477350-03-1 477350-04-2

477350-05-3 477350-06-4 477350-07-5 477350-08-6

477350-09-7 477350-10-0, 5,6-Epoxy-3,3,5-trimethyl-1-hexene

477350-11-1 477350-12-2 477350-13-3

RL: CAT (Catalyst use); USES (Uses)

(inimer; arborescent polyolefinic thermoplastic elastomers and

products therefrom)

REFERENCE COUNT:

THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L44 ANSWER 7 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2002:742328 HCAPLUS Full-text

DOCUMENT NUMBER: 138 - 55759

TITLE: Combined Lewis acid

catalysts in shotgun process: a

convenient synthesis of the female sex pheromone

of the red-bollworm moth

AUTHOR(S): Nagano, Yoshifumi; Orita, Akihiro; Otera, Junzo CORPORATE SOURCE:

Department of Applied Chemistry, Okayama

University of Science, Ridai-cho, Okayama, 700-0005, Japan

SOURCE: Tetrahedron (2002), 58(41), 8211-8217

CODEN: TETRAB; ISSN: 0040-4020

PUBLISHER: Elsevier Science Ltd.

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 138:55759

AB The combined use of Lewis acid and distannoxane catalysts gives rise to a new variant of the shotgun process. The unwanted acetylation of a secondary homoallyl alc. by the former catalyst is suppressed through hybridization with the latter resulting in one-pot aldehyde allylation and primary alc.

acetylation of ω -hydroxy alkanal without protection/deprotection procedures. IT 91743-86-IP

RL: SPN (Synthetic preparation); PRBP (Preparation)
(a shotgun one-pot process catalyzed by Lewis
acids and distannoxane for aldehyde allylation and
primary alc. acetylation of a @-hydroxy alkanal without
protection/deprotection procedures, and its application to the
synthesis of a pheromone)

RN 91743-86-1 HCAPLUS

CN 1-Dodecen-4-ol, 4-acetate (CA INDEX NAME)

CC 26-2 (Biomolecules and Their Synthetic Analogs)
ST pheromone synthesis; Lewis acid distannoxane

combined catalyst acetylation allylation pheromone synthesis; shotoun one pot allylation acetylation process

IT Acetylation Allylation

(a shotgun one-pot process catalyzed by Lewis acids and distannoxane for aldehyde allylation and primary alc. acetylation of a on-hydroxy alkanal without protection/deprotection procedures, and its application to the synthesis of a pheromone)

IT Lewis acids

RL: CAT (Catalyst use); USES (Uses)

(a shotgun one-pot process catalyzed by Lewis acids and distannoxane for aldehyde allylation and

primary alc. acetylation of a ω -hydroxy alkanal without protection/deprotection procedures, and its application to the synthesis of a pheromone)

IT 109-63-7, Boron trifluoride etherate 27607-77-8, Trimethylsilyl triflate 33194-92-2 34946-82-2 88189-03-1, Bismuth triflate 144026-79-9, Scandium triflate

RL: CAT (Catalyst use); USES (Uses)

(a shotgun one-pot process catalyzed by Lewis acids and distannoxane for aldehyde allylation and

primary alc. acetylation of a ω -hydroxy alkanal without protection/deprotection procedures, and its application to the synthesis of a pheromone)

IT 111-87-5, 1-Octanol, reactions 124-19-6, Nonanal 7393-43-3, Tetraallyl tin 13019-22-2, 9-Decen-1-ol

RL: RCT (Reactant); RACT (Reactant or reagent)
(a shotgun one-pot process catalyzed by Lewis

acids and distannoxane for aldehyde allylation and

primary alc. acetylation of a ω -hydroxy alkanal without protection/deprotection procedures, and its application to the

synthesis of a pheromone) 22054-15-5P 124388-99-4P 479206-93-4P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation);

RACT (Reactant or reagent) (a shotgun one-pot process catalyzed by Lewis acids and distannoxane for aldehyde allylation and

primary alc. acetylation of a @-hydroxy alkanal without protection/deprotection procedures, and its application to the

synthesis of a pheromone) 112-14-1P 50767-78-7P 77383-04-1P, 1-Dodecen-4-ol

91743-86-1P 479206-88-7P 479206-89-8P. 11-Dodecene-1,9-dio1 479206-90-1P 479206-91-2P 479206-92-3P

479206-94-5P RL: SPN (Synthetic preparation); PREP (Preparation) (a shotgun one-pot process catalyzed by Lewis

acids and distannoxane for aldehyde allylation and primary alc. acetylation of a @-hydroxy alkanal without protection/deprotection procedures, and its application to the synthesis of a pheromone)

REFERENCE COUNT: 31

THERE ARE 31 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L44 ANSWER 8 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2000:776693 HCAPLUS Full-text

DOCUMENT NUMBER:

134:71936 Living coupling reaction in living cationic

TITLE:

polymerization. 4. Synthesis of telechelic

polyisobutylenes using bis-furanyl derivatives as coupling agents

Hadjikvriacou, Savvas; Faust, Rudolf; Suzuki,

AUTHOR(S): Toshio

CORPORATE SOURCE:

Polymer Science Program Department of Chemistry, University of Massachusetts-Lowell, Lowell, MA,

01854, USA

SOURCE: Journal of Macromolecular Science, Pure and Applied Chemistry (2000), A37(11), 1333-1352

CODEN: JSPCE6; ISSN: 1060-1325

Marcel Dekker, Inc.

DOCUMENT TYPE: Journal

LANGUAGE: English

PUBLISHER:

The synthesis of telechelic polvisobutylenes (PIB) by coupling with 2,5-bis-(2-fury1-2-propy1) furan is described. Hydroxyl telechelic PIB with controlled mol. weight was obtained by haloboration-initiation of isobutylene (IB) with a BC13/BBr3 mixture, followed by coupling of the living ends and oxidation of the (CH3O)2B- end groups with H2O2 in alkaline THF. Vinvl telechelic PIBs were prepared by coupling living PIB obtained using a novel vinyl functional initiator 5-chloro-3,3,5-trimethyl-1-hexene in conjunction with TiCl4 in hexanes/CH3Cl (60/40, volume/volume) at -80°C. Chlorosilyl telechelic PIBs were synthesized for the first time by employing chlorosilyl functional initiators

(1-chloro-1-methyl)ethyl-3-(1-dichloro-methylsilyl- methyl)ethylbenzene and 1methyldichlorosilv1-3,3,5-trimethyl-5- chloro-hexane in conjunction with TiCl4 in hexanes/CH3Cl (60/40, volume/volume) at -80°C. After coupling, the dichlorosilyl groups were converted to dimethoxysilyl groups upon guenching with methanol. The bis-dimethoxysilyl telechelic PIBs were crosslinked by moisture at room temperature in the presence of catalytic amts. of tin(II) 2ethyl-hexanoate. Extraction of the crosslinked samples with hexanes resulted in negligible soluble content, indicating essentially quant. crosslinking.

69078-25-7P, 5-Chloro-3,3,5-trimethyl-1-hexene

RL: CAT (Catalyst use); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent); USES (Usea)

(polymerization catalyst; synthesis of telechelic polyisobutylenes using bis-furanyl derivs. as coupling agents)

RN 69078-25-7 HCAPLUS

CN 1-Hexene, 5-chloro-3,3,5-trimethyl- (CA INDEX NAME)

$$\begin{array}{c} \texttt{C1} & \texttt{Me} \\ \texttt{Me} - \overset{\texttt{C}}{\overset{\texttt{C}}}{\overset{\texttt{C}}{\overset{\texttt{C}}}{\overset{\texttt{C}}{\overset{\texttt{C}}}{\overset{\texttt{C}}{\overset{\texttt{C}}}{\overset{\texttt{C}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}}{\overset{\texttt{C}}}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}}{\overset{\texttt{C}}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{C}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}}{\overset{\texttt{C}}}}{\overset{C}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{\texttt{C}}}}{\overset{$$

CC 35-3 (Chemistry of Synthetic High Polymers)

IT 69078-25-7P, 5-Chloro-3,3,5-trimethyl-1-hexene

RL: CAT (Catalyst use); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent); USES

(Uses)

(polymerization catalyst; synthesis of telechelic polyisobutylenes using bis-furanyl derivs. as coupling agents)

IN THE RE FORMAT

REFERENCE COUNT: 21 THERE ARE 21 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE

L44 ANSWER 9 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1999:719035 HCAPLUS Full-text

DOCUMENT NUMBER: 131:337525

TITLE: Silyl-functional initiator for living cationic

polymerization

INVENTOR(S): Faust, Rudolf; Hadjikyriacou, Savvas E.; Suzuki,

Toshio
PATENT ASSIGNEE(S): Universi

PATENT ASSIGNEE(S): University of Massachusetts, USA; Dow Corning Corporation

SOURCE: U.S., 7 pp.
CODEN: USXXAM

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PAT	ENT I	NO.			KIND DATE				APPLICATION NO.						ATE	
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US	5981	/85			A		1999	1109		JS I	998-	2049	00		1:	99812
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		IN,	IS,	JP,	KE,	KG,	KP,	KR,	KZ,	LC,	LK,	LR,	LS,	LT,	LU,	LV,
		MD,	MG,	MK,	MN,	MW,	MX,	NO,	NZ,	PL,	PT,	RO,	RU,	SD,	SE,	SG,
		SI,	SK,	SL,	TJ,	TM,	TR,	TT,	UA,	UG,	UZ,	VN,	YU,	ZA,	ZW,	AM,
		AZ,	BY,	KG,	KZ,	MD,	RU,	TJ,	TM							
	RW:	GH,	GM,	KE,	LS,	MW,	SD,	SL,	SZ,	TZ,	UG,	ZW,	AT,	BE,	CH,	CY,
		DE,	DK,	ES,	FΙ,	FR,	GB,	GR,	IE,	IT,	LU,	MC,	NL,	PT,	SE,	BF,

BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG PRIORITY APPLN. INFO.: US 1998-204900 A 199812

0.3

OTHER SOURCE(S): MARPAT 131:337525

Haloorganosilane initiators which can be reacted with cationically polymerizable monomers in the presence of a Lewis acid to provide living polymers having a silvl-functional head group have the structure XaR3aSiZCR2R1, where X is halogen, R is independently selected from alkyl groups having 1-10 carbon atoms or aryl groups having 6-10 carbon atoms, Z is a divalent aliphatic hydrocarbon group having at least 3 carbon atoms, R1 is halogen, alkoxy, acyloxy or hydroxy, and a is 1, 2, or 3. Thus, 3,3,5trimethyl-5-chloro-1-hexene was treated with dichloromethylsilane in the presence of a platinum-divinyltetramethyldisiloxane complex catalyst to give a hydrosilated product which was rehydrochlorinated by treatment with HCl to give 1-methyldichlorosily1-3,3,5-trimethy1-5- chlorohexane (I). Isobutylene was polymerized in a MeCl/hexane solvent system using I, titanium tetrachloride, and 2,6-di-tert-butylpyridine as catalysts. The resulting cationic living polymer having number-average mol. weight 3200 was coupled using 2,5-bis(2-furylpropyl)furan as the coupling agent to give a polymer which on treatment with MeOH gave a methoxysilyl-functional telechelic polvisobutylene (II) having number-average mol. weight 5300. II was cured by exposure to moisture in the presence of tin octoate and could be used in the formulation of sealants, adhesives and coatings,

ΙT 69078-25-7

RL: RCT (Reactant); RACT (Reactant or reagent)

(reactant; in preparation of haloorganosilane initiators for living cationic polymerization of vinyl monomers)

RN 69078-25-7 HCAPLUS

CN 1-Hexene, 5-chloro-3,3,5-trimethyl- (CA INDEX NAME)

TC TCM C07F007-12

INCL 556488000

35-3 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 39

haloorganosilane initiator living cationic polymn; isobutylene living cationic polymn silane catalyst; moisture curable polyisobutylene

IT Polymerization

Polymerization catalysts

(cationic, living; reactive silvl-functional initiators for living cationic polymerization of vinyl monomers and preparation of

polymers containing hydrolyzable silvl end groups)

ΙT Hydrosilylation catalysts

> (platinum organosiloxane complex; for preparation of silvl-functional initiators for living cationic polymerization of vinyl compds.)

7440-06-4D, Platinum, complex with divinyltetramethyldisiloxane, uses 30110-75-9D, Divinyltetramethyldisiloxane, platinum complex RL: CAT (Catalyst use); USES (Uses)

(hydrosilylation catalyst; for preparation of

silyl-functional initiators for living cationic polymerization of vinyl compds.)

IT 249905-14-4P

RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)

(polymerization catalyst; in polymerization of isobutylene)

IT 249905-12-2P

RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP

(Preparation); USES (Uses)

(polymerization catalyst; reactive silyl-functional initiators

for living cationic polymerization of vinyl monomers and preparation of polymers containing hydrolyzable silvl end groups)

IT 75-54-7, Dichloromethylsilane 78-79-5, Isoprene, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(reactant; in preparation of haloorganosilane initiators for living cationic polymerization of vinyl monomers)

cationic polymerization of vinyl monor

REFERENCE COUNT: 14 THERE ARE 14 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L44 ANSWER 10 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1998:801484 HCAPLUS Full-text

DOCUMENT NUMBER: 130:167901

TITLE: Relative reactivities of acetals and ethers

under Friedel-Crafts conditions

AUTHOR(S): Mayr, Herbert; Dau-Schmidt, Jan-Peter

CORPORATE SOURCE: Institut fur Organische Chemie, Technische

Hochschule Darmstadt, Darmstadt, 64287, Germany SOURCE: Chemische Berichte (1994), 127(1), 213-217

CODEN: CHBEAM; ISSN: 0009-2940

VCH

PUBLISHER: VCH
DOCUMENT TYPE: Journal
LANGUAGE: English

AB Competition expts. have been performed to determine the relative reactivities of acetals and ethers toward allyltrimethylsilane in the presence of catalytic amts. of BF3-0Et2. It is found that acetals R-CH(OMe)2 and their phenylogous p-anisyl ethers R-CH(p-MeOC6H4) (OMe) show very little differences in reactivity. The reactivity scales are employed to rationalize the results of Lewis acid-catalyzed addns. of acetals and ethers to CC double bonds.

IT 54267-82-2P, 4-Methoxy-1-heptene

RL: SPN (Synthetic preparation); PREP (Preparation)

(relative reactivities of acetals and ethers under Friedel-Crafts conditions)

RN 54267-82-2 HCAPLUS

CN 1-Heptene, 4-methoxy- (CA INDEX NAME)

CC 22-4 (Physical Organic Chemistry)

IT Friedel-Crafts reaction

Friedel-Crafts reaction catalysts

(relative reactivities of acetals and ethers under Friedel-Crafts conditions)

IT 109-63-7, Boron trifluoride etherate 21512-92-5 RL: CAT (Catalyst use); USES (Uses) 71104-84-2P

(catalyst; relative reactivities of acetals and ethers under Friedel-Crafts conditions)

111874-57-8P 125310-48-7P 220331-49-7P 220331-50-0P RL: SPN (Synthetic preparation); PREP (Preparation)

(relative reactivities of acetals and ethers under Friedel-Crafts conditions)

REFERENCE COUNT: 14 THERE ARE 14 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L44 ANSWER 11 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1997:714668 HCAPLUS Full-text DOCUMENT NUMBER: 128:12964

ORIGINAL REFERENCE NO.: 128:2513a,2516a

54267-82-2P, 4-Methoxy-1-heptene

TITLE: Bidentate organoaluminum Lewis acid for selective activation of carbonyl over acetal functionality: chemoselective functionalization

AUTHOR(S): Ooi, Takashi; Tayama, Eiji; Takahashi, Makoto;

Maruoka, Keiji

Department of Chemistry, Graduate School of CORPORATE SOURCE: Science, Hokkaido University, Sapporo, 060,

Japan

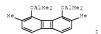
Tetrahedron Letters (1997), 38(42), 7403-7406 SOURCE:

CODEN: TELEAY; ISSN: 0040-4039

PUBLISHER: Elsevier DOCUMENT TYPE: Journal LANGUAGE: English

CASREACT 128:12964 OTHER SOURCE(S):

GI



- AB Chemoselective functionalization of carbonyl compds, over acetals has been achieved by bidentate organoaluminum Lewis acid based on the selective double electrophilic activation of carbonyls. E.g., Mukaiyama aldol reaction of PhCHO and PhCH(OMe)2 with 1-(trimethylsiloxy)-1-cyclohexene in presence of organoaluminum Lewis acid I gave predominantly the product resulting from reaction of PhCHO.
- 153864-63-2P

RL: SPN (Synthetic preparation); PREP (Preparation) (bidentate organoaluminum Lewis acid for selective activation of carbonyl over acetal groups)

RN 153864-63-2 HCAPLUS

1-Decene, 4-methoxv- (CA INDEX NAME)

H2C CH_ CH2_CH_ (CH2)5_Me

```
21-2 (General Organic Chemistry)
    organoaluminum Lewis acid catalyst
    carbonyl reaction
    Aldol condensation
ΙT
    Aldol condensation catalysts
        (Mukaiyama; bidentate organoaluminum Lewis acid
        for selective activation of carbonyl over acetal groups)
    Acetals
     Carbonvl compounds (organic), reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (bidentate organoaluminum Lewis acid for
        selective activation of carbonyl over acetal groups)
     184020-59-5P
     RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP
     (Preparation); USES (Uses)
        (bidentate organoaluminum Lewis acid for
        selective activation of carbonyl over acetal groups)
    75-24-1, Trimethylaluminum 98-86-2, Acetophenone, reactions
     100-52-7, Benzaldehyde, reactions 112-12-9, Methyl nonyl ketone
     124-13-0, Octanal 1125-88-8, Benzaldehyde dimethyl acetal
     2043-61-0, Cyclohexanecarboxaldehyde 4316-35-2, Acetophenone
     dimethyl acetal 6651-36-1, 1-(Trimethylsiloxy)-1-cyclohexene
     10022-28-3 18231-08-8, Cyclohexanecarboxaldehyde dimethyl acetal
     24850-33-7, Allyltributyltin 31469-15-5
                                                68298-50-0 98991-01-6
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (bidentate organoaluminum Lewis acid for
        selective activation of carbonyl over acetal groups)
     936-58-3P 22039-97-0P 35022-33-4P 36971-14-9P, 1-Decen-4-ol
     54322-88-2P 56072-25-4P 61841-05-2P 85670-57-1P 94123-64-5P 113812-36-5P 153864-63-2P 175732-06-6P 198979-20-3P
     198979-21-4P 198979-22-5P 198979-23-6P
                                                 198979-24-7P
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (bidentate organoaluminum Lewis acid for
       selective activation of carbonyl over acetal groups)
                              THERE ARE 15 CITED REFERENCES AVAILABLE
REFERENCE COUNT:
                         15
                               FOR THIS RECORD. ALL CITATIONS AVAILABLE
                               IN THE RE FORMAT
L44 ANSWER 12 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER: 1989:155017 HCAPLUS Full-text
DOCUMENT NUMBER:
                        110:155017
ORIGINAL REFERENCE NO.: 110:25659a,25662a
TITLE:
                        Initiators and process for living polymerization
                        of olefins to end-functionalized polymers
INVENTOR(S):
                        Kennedy, Joseph Paul; Mishra, Munmaya Kumar
PATENT ASSIGNEE(S):
SOURCE:
                        Eur. Pat. Appl., 34 pp.
                        CODEN: EPXXDW
DOCUMENT TYPE:
                        Patent
LANGUAGE:
                        English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
```

DATENT NO	KIND	DATE	APPLICATION NO.	DATE
PATENT NO.	KIND		APPLICATION NO.	DATE
PR 055050		40000400	PP 4000 200000	
EP 265053	A2	19880427	EP 1987-307508	
				198708

	, =000				10,000,,,,,				
	265053 265053			В1	19900829 19961023				
		BE,	CH,		FR, GB, IT,				
CA	1338520			С	19960813	CA	1987-545335		
									198708 25
AT	144542			T	19961115	AT	1987-307508		
									198708 25
JP	01062308			A	19890308	JP	1987-216542		
									198709 01
JP	2707086			B2	19980128				
US	4929683			Α	19900529	US	1988-173049		
									198803 28
US	5066730			A	19911119	US	1990-500771		
									199003 28
PRIORIT	Y APPLN. :	INFO.	:			US	1986-899655	A	
									198608 25
						IIS	1988-173049	A3	
						05	1,000 173049	AS	198803 28

AB The title living vinvl polymers, having specific end groups and a wide range of mol. wts., are prepared by cationic living polymerization of vinyl monomers [e.g., isobutylene (I) or styrene] in the presence of complexes of organic ethers and Lewis acids comprising BC13, AlC13, SnC14, TiC14, SbF5, FeC13, ZnCl2, and VCl4. Cumyl Me ether (II) 7.7 + 10-2, I 1.17, and BCl3 0.37 mol/L were stirred in 20 mL CH2Cl2 at -30° for 0.5 h to give a I polymer terminated with C6H5C(CH3)2 and C1 and having number-average mol. weight (Mn) 1500, weight-average mol. weight Mw 5000, and Mn/Mw 3.3. A polymer having Mn 13,700; Mw 30,000, and Mn/Mw 2.2 was obtained when 7.4 + 10-3 mol/L II was used.

119886-65-6 ΙT

RL: USES (Uses)

(initiator, for preparation of cationic living polymer systems)

RN 119886-65-6 HCAPLUS

CN Boron, trichloro(4-methoxy-1-pentene)-, (T-4)- (CA INDEX NAME)

. Н— СН 2— СН—— СН 2

IC ICM C08F010-00 ICS C08F004-06

35-3 (Chemistry of Synthetic High Polymers) Section cross-reference(s): 67

ether trichloroboron complex polyisobutylene initiator; boron trichloro ether complex initiator; cationic polymn Lewis acid ether complex; chlorine terminated polyisobutylene; catalyst living polymn isobutylene; styrene living polymn

catalyst; mol wt control polvisobutylene catalyst; polydispersity polyisobutylene ether Lewis acid; chain transfer agent polyisobutylene тт Polymerization catalysts (ether-Lewis acid compds., for preparation of living cationic polymer systems) Polymerization (cationic, living, of olefins, ether-Lewis acid compds. for) Ethers, compounds RL: USES (Uses) (compds., with Lewis acids, initiators, for living cationic polymerization of olefins) lewis anids RL: USES (Uses) (compds., with ethers, initiators, for living cationic polymerization of olefins) Polymers, preparation RL: IMF (Industrial manufacture); PREP (Preparation) (telechelic, manufacture of, having low polydispersity, Lewis acid-ether compound initiators for) 58286-97-8 119865-82-6 119865-83-7 119865-84-8 119886-47-4 119886-48-5 119886-49-6 119886-50-9 119886-51-0 119886-52-1 119886-53-2 119886-54-3 119886-55-4 119886-56-5 119886-57-6 119886-58-7 119886-59-8 119886-60-1 119886-61-2 119886-62-3 119886-63-4 119886-64-5 119886-65-6 119886-66-7 119886-67-8 119886-68-9 119886-69-0 119886-70-3 119886-71-4 119886-72-5 RL: USES (Uses) (initiator, for preparation of cationic living polymer systems) 9010-85-9P RL: IMF (Industrial manufacture); PREP (Preparation) (manufacture of, with low polydispersity, Lewis acid -ether compound polymerization initiators for) 9003-27-4DP, Polvisobutylene, chlorine and tertiary hydrocarbyl-terminated 9003-53-6DP, Polystyrene, chlorine and tertiary hydrocarbyl-terminated RL: IMF (Industrial manufacture); PREP (Preparation) (manufacture of, with low polydispersity, Lewis acid -ether complex polymerization initiators for) L44 ANSWER 13 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1988:509927 HCAPLUS Full-text DOCUMENT NUMBER: 109:109927 ORIGINAL REFERENCE NO.: 109:18303a,18306a TITLE: Methoxy(phenylthio)methane as an ambi-equivalent of a methoxy- or (phenylthio)methylene 1,1-dipole AUTHOR(S): Sato, Tsuneo; Okura, Shuji; Otera, Junzo; Nozaki, Hitoshi CORPORATE SOURCE: Dep. Appl. Chem., Okayama Univ. Sci., Okayama, 700, Japan Tetrahedron Letters (1987), 28(50), 6299-302 SOURCE: CODEN: TELEAY: ISSN: 0040-4039 DOCUMENT TYPE: Journal LANGUAGE: English OTHER SOURCE(S): CASREACT 109:109927 MeOCHRSPh [R = (CH2)8H, (CH2)8OSiMe2CMe3, CH2CH:CMe2, Ph], prepared by alkylation of MeOCH2SPh, are allylated by R1CR2:CHR3CH2SnBu3 (R1-R3 = H; R1 =

R2 = H, R3 = Me; R1 = R2 = Me, R3 = H; R1 = Me, R2 = R3 = H) in the presence

of BF3.OEt2 to give almost entirely MeOCHRCR1R2CR3:CH2. In the presence of TiCl4, however, the product is almost entirely PhSCHRCR1R2CR3:CH2.

116204-91-2P, 4-Methoxy-1-dodecene 116204-93-4P, 4-Methoxy-2-methyl-1-dodecene

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

RN 116204-91-2 HCAPLUS

1-Dodecene, 4-methoxy- (CA INDEX NAME) CN

116204-93-4 HCAPLUS RN

CN 1-Dodecene, 4-methoxy-2-methyl- (CA INDEX NAME)

- 25-9 (Benzene, Its Derivatives, and Condensed Benzenoid Compounds) Section cross-reference(s): 23
- ST allylation methoxyphenylthioalkane titanium catalyst regiochem; allyltin allylation methoxyphenylthioalkane regiochem; boron trifluoride catalyst allylation methoxyphenythioalkane; homoallyl methyl ether; phenyl homoallyl

sulfide; alkene methoxy; phenylthioalkene

Regiochemistry

(of allylation of methoxy(phenylthio)alkanes with allyltin reagents, Lewis acid-promoted)

Allylation

(regioselective, of methoxy(phenylthio)alkanes with allyltin reagents, Lewis acid-promoted)

7695-69-4, Benzaldehyde bis(phenylthio)acetal

RL: RCT (Reactant); RACT (Reactant or reagent)

(allylation of, with allyltributyltin, Lewis

acid-promoted regioselective)

762-72-1, Allyl trimethylsilane

RL: RCT (Reactant); RACT (Reactant or reagent)

(allylation of, with methoxy(phenylthio)nonane, Lewis

acid-promoted regioselective)

35998-93-7 53911-92-5

RL: RCT (Reactant); RACT (Reactant or reagent)

(allylation with, of methoxy(phenylthio)alkane, Lewis acid-promoted regioselective)

24850-33-7, Allyl tributyltin

RL: RCT (Reactant); RACT (Reactant or reagent)

(allylation with, of methoxy(phenylthio)alkanes, Lewis

acid-promoted regioselective)

67883-62-9

RL: RCT (Reactant); RACT (Reactant or reagent)

(allylation with, of methoxy(phenylthio)nonane, Lewis

acid-promoted regioselective)

99054-47-4P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation and allylation and propargylation of, with tin reagents, Lewis acid-promoted regioselective)

13865-07-1P 116204-90-1P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation);

RACT (Reactant or reagent)

(preparation and allylation of, with allyltin reagents, Lewis acid-promoted regioselective)

116204-89-8P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and allylation of, with allyltributyltin, Lewis acid-promoted regioselective)

21213-27-4P 22039-97-0P 50767-78-7P.

(E)-1-Acetoxy-9,11-dodecadiene 54905-16-7P 92737-75-2P 92737-76-3P 116204-91-2P, 4-Methoxy-1-dodecene

116204-92-3P, 4-(Phenylthio)-1-dodecene 116204-93-4P,

4-Methoxy-2-methyl-1-dodecene 116204-94-5P,

2-Methyl-4-(phenylthio)-1-dodecene 116204-95-6P,

4-Methoxy-1-dodecyne 116204-96-7P, 4-(Phenylthio)-1-dodecyne

116204-97-8P 116204-99-0P, 5-Methoxy-2-methyl-2,7-octadiene

116205-00-6P, 2-Methyl-5-(phenylthio)-2,7-octadiene 116205-01-7P,

4-Methoxy-3,3,7-trimethyl-1,6-octadiene 116205-02-8P, 4-(Phenylthio)-3,3,7-trimethyl-1,6-octadiene 116205-03-9P

116205-04-0P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

53915-69-8, Allenyl tributyltin RL: RCT (Reactant); RACT (Reactant or reagent)

(propargylation with, of methoxy(phenylthio)nonane, Lewis acid-promoted regioselective)

L44 ANSWER 14 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1988:111834 HCAPLUS Full-text

DOCUMENT NUMBER: 108:111834

ORIGINAL REFERENCE NO.: 108:18297a,18300a

TITLE: Lewis acid-induced reaction

of γ , δ -epoxy tin compounds

Sato, Tadashi; Watanabe, Masami; Murayama, AUTHOR(S):

Eigoro Dep. Appl. Chem., Waseda Univ., Tokyo, 160,

Japan

SOURCE: Synthetic Communications (1987), 17(7), 781-8

CODEN: SYNCAV; ISSN: 0039-7911

Journal DOCUMENT TYPE:

LANGUAGE: English

OTHER SOURCE(S): CASREACT 108:111834

AB (Stannylethyl)oxiranes I (R1 = alkyl, H; R2 = H, Me; R3 = H, Me) were treated with BF3.Et20 in CH2C12 to give cyclopropanemethanols II.

113235-26-0P

CORPORATE SOURCE:

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

RN 113235-26-0 HCAPLUS

CN 1-Nonene, 4-chloro- (CA INDEX NAME)

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\begin{array}{c} \text{C1} \\ \text{H2C} \longrightarrow \text{CH} \longrightarrow \text{CH2} \\ \text{H} \longrightarrow \text{CH2} \\ \text{OH2} \\ \text{J} \longrightarrow \text{Me} \end{array}
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CC 24-2 (Alicyclic Compounds)

Section cross-reference(s): 27, 29

ST cyclopropanemethanol alkyl; oxirane stannylethyl rearrangement catalyst; rearrangement stannylethyloxirane catalyst

IT Rearrangement catalysts

(boron fluoride etherate and titanium tetrafluoride, for

[(trimethylstannyl)ethyl]oxiranes) II 109-63-7, Boron trifluoride etherate 7550-45-0, uses and

miscellaneous

RL: CAT (Catalyst use); USES (Uses) (catalyst, for rearrangement of

[(trimethylstannyl)ethyl]oxiranes)

IT 113235-13-5P

RL: RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)

(formation and rearrangement of, catalyst for)

II 113235-11-3P 113235-12-4P 113235-14-6P 113235-15-7P
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation);
RACT (Reactant or reagent)

(preparation and rearrangement of, catalysts for)

IT 2746-14-7P 21003-35-0P 21003-36-1P 26001-59-2P 113235-16-8P 113235-17-9P 113235-18-0P 113235-22-6P 113235-23-7P 113235-24-8P 113235-25-9P 113235-26-0P 113235-27-1P

113301-18-1P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

T 113235-19-1 113235-20-4 113235-21-5

RL: RCT (Reactant); RACT (Reactant or reagent) (rearrangement of, by (phenylthio)silane derivative,

catalysts for)

L44 ANSWER 15 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1983:438666 HCAPLUS Full-text

DOCUMENT NUMBER: 99:38666

ORIGINAL REFERENCE NO.: 99:6085a,6088a

TITLE: Chemistry of organosilicon compounds - 165.
2-(Trimethylsilylmethyl)-1,3-butadiene - a

versatile building block for terpene synthesis
AUTHOR(S): Sakurai, Hideki; Hosomi, Akira; Saito, Masaki;

Sasaki, Koshi; Iguchi, Hirokazu; Sasaki, Junichi; Araki, Yoshitaka

CORPORATE SOURCE: Dep. Chem., Tohoku Univ., Sendai, 980, Japan

SOURCE: Tetrahedron (1983), 39(6), 883-94 CODEN: TETRAB; ISSN: 0040-4020

DOCUMENT TYPE: Journal

LANGUAGE: English

AB Me3SiGH2C(:CH2)CH:CH2, prepared by Grignard coupling of Me3SiGH2C1 with CH2:C(C1)CH:CH2, isoprenylated acid chlorides, aldehydes, ketones, and acetals in the presence of Lewis acids and underwent Diels-Alder reaction with

dienophiles with high regiospecificity to give synthons for limonene, $\delta\text{-terpineol},$ isobisabolene, etc.

IT 75243-70-8P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

RN 75243-70-8 HCAPLUS

CN 1-Octene, 5-methoxy-7-methyl-3-methylene- (CA INDEX NAME)

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i-Bu-CH-CH2-CH2
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CC 30-15 (Terpenes and Terpenoids)
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IT 14264-16-5 14647-23-5 15629-92-2 RL: CAT (Catalyst use); USES (Uses)

(catalysts, for coupling of silylmethylmagnesium

chloride with chloroprene)

539-70-8P 6090-09-1P 6493-78-3P 6493-79-4P 14314-21-7P 14434-41-4P 17023-58-4P 19860-68-5P 41723-53-9P 70901-65-4P 70901-69-8P 70901-71-2P 70901-68-7P 70901-72-3P 71092-48-3P 71092-49-4P 71092-52-9P 74043-01-9P 71092-54-1P 74043-02-0P 74043-04-2P 74043-05-3P 74043-06-4P 74043-07-5P 74043-10-0P 74043-11-1P 74043-12-2P 74043-13-3P 74043-14-4P 74043-15-5P 74043-16-6P 75243-70-8P 75243-71-9P 82096-09-1P 82096-10-4P 82096-12-6P 82096-13-7P 86361-18-4P 86361-19-5P 86361-20-8P 86361-23-1P 86361-25-3P 86361-44-6P 86372-75-0P 86372-76-1P 86372-77-2P 86372-78-3P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

L44 ANSWER 16 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1979:457185 HCAPLUS Full-text

DOCUMENT NUMBER: 91:57185 ORIGINAL REFERENCE NO.: 91:9275a,9278a

TITLE: Chemistry of organosilicon compounds. 120.

2-Trimethylsilylmethyl-1,3-butadiene as a novel reagent for isoprenylation. New access to ipsenol and ipsdienol, pheromones of Ips

paraconfusus

AUTHOR(S): Hosomi, Akira; Saito, Masaki; Sakurai, Hideki CORPORATE SOURCE: Dep. Chem., Tohoku Univ., Sendai, Japan SOURCE: Tetrahedron Letters (1979), (5), 429-32

CODEN: TELEAY; ISSN: 0040-4039

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 91:57185

AB CH2:C(CH2S1Me3)CH:CH2 (I) was prepared (91%) from CH2:CC1CH:CH2 by treatment with Me3SiCH2MgCl in Et2O (reflux, 6 h) in the presence of Ni[Ph2P(CH2)3PPh2]Cl2 followed by hydrolysis. I reacted with aliphatic and aromatic acetals, acid chlorides, carbonyl compds., and other electrophiles in the presence of a Lewis acid to qive isoprenylated products. E.g., I on sequential treatment with Me2CHCH2COCl (TiCl4, CH2Cl2) and (Me2CHCH2)2AlH (C6H6) gave 62% (t)-ipsenol. I with Me2C:CHCOCl (TiCl4, CH2Cl2, 10 min) gave 71% myrcenone which on similar reduction gave 75% (t)-ipsdienol.

T 75243-70-8P 75243-73-1P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

RN 75243-70-8 HCAPLUS

32

1-Octene, 5-methoxy-7-methyl-3-methylene- (CA INDEX NAME)

RN 75243-73-1 HCAPLUS

CN 1-Nonene, 5-ethoxy-3-methylene- (CA INDEX NAME)

30-10 (Terpenoids)

Section cross-reference(s): 5, 23, 25, 29

Grignard reaction catalysts

(nickel dichloro phosphines, for (chloromethyl)trimethylsilane with chloroprene)

TT Alkylation catalysts

(isoprenylation, Lewis acids, for

electrophiles by isoprenvlsilane)

14264-16-5 14647-23-5 15629-92-2 RL: CAT (Catalyst use); USES (Uses)

(catalyst, for Grignard reaction of

(chloromethyl)trimethylsilane with chlorobutadiene)

7446-70-0, uses and miscellaneous 7550-45-0, uses and miscellaneous

RL: CAT (Catalyst use); USES (Uses)

(catalyst, for isoprenylation of electrophiles by

isoprenylsilane) 70901-65-4P 70901-68-7P 70901-69-8P 70901-71-2P 70901-72-3P

75243-70-8P 75243-71-9P 75243-73-1P RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of)

L44 ANSWER 17 OF 17 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1965:446640 HCAPLUS Full-text

DOCUMENT NUMBER: 63:46640

ORIGINAL REFERENCE NO.: 63:8497h,8498a-b

TITLE: Copolymerization of α -olefins with

 ω -halo α -olefins by use of Ziegler

catalysts

AUTHOR(S): Bacskai, R. CORPORATE SOURCE:

California Res. Corp., Richmond SOURCE:

Journal of Polymer Science, Part A: General

Papers (1965), 3(7), 2491-510 CODEN: JPYAAK; ISSN: 0449-2951

DOCUMENT TYPE: Journal

LANGUAGE: English

AR The copolymerization of α -olefins with 5-chloro-1-pentene, 6-chloro-1-hexene, 8-chloro-1-octene, 7-chloro-1-octene, 3-bromo-1-octene, 4-bromo-1-butene, 5bromo-1-pentene, 6-bromo-1-hexene, and 11-chloro-1-undecene by use of Ziegler catalysts has been investigated. w-Halo a-olefins deactivate the catalyst, thus resulting in low copolymer yield and reduced mol. weight The deactivation is caused by H halide formation which is greatly facilitated in

the case of the allyl halides and 6-halo-1-hexenes. The different degrees of deactivation were interpreted by considering the ease of carbonium ion formation from the various ω -halo α -olefins. The copolymer yields were considerably improved when a catalytic amount of a pyridine, Et3N, or tetrahydrofuran was added to the polymerization system. It is proposed that the added Lewis base, by decreasing the Lewis acid character of the catalyst, suppresses the decomposition of the ω -halo α -olefin leading to B halide formation. Depending on the Ziegler catalyst employed, the copolymers of α -olefins and ω -halo α -olefins can be either crystalline or amorphous solids. The halogen, especially the Br in the copolymers, is reactive in nucleophilic substitution reactions. Reaction with amines gave polymers which could be dyed by acid dyes.

IT 31950-56-8

(Derived from data in the 7th Collective Formula Index (1962-1966))

RN 31950-56-8 HCAPLUS

CN 1-Pentene, 4-bromo- (CA INDEX NAME)

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Br
Me—CH—CH2—CH—CH2
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CC 45 (Synthetic High Polymers)

IT Bases

(Lewis, catalysts from Al compds., TiCl3 and, in polymerization of ω -halo α -olefins with α -olefins)

α-olerins)
Catalysts and Catalysis

(in polymerization, of ω-halo α-olefins with

α-olefins, Ziegler)
T Polymerization

(of ω -halo α -olefins with α -olefins, Ziegler catalysts in)

IT Olefins

(polymerization of α -, with ω -halo α -olefins, Ziegler catalysts in)

IT Olefins

(ω -halo α -, polymerization with α -olefins, Ziegler catalysts in)

T 4984-01-4 5026-95-9 31950-56-8 95347-52-7

(Derived from data in the 7th Collective Formula Index (1962-1966))

IT 7705-07-9, Titanium chloride, TiCl3

(catalysts from Al compds. and TiCl3 in polymerization, of $\omega\text{-halo}~\alpha\text{-olefins}$ with $\alpha\text{-olefins})$

II 109-99-9, Furan, tetrahydro- 110-86-1, Pyridine 121-44-8, Triethylamine

(catalysts from Al compds., TiCl3 and, in polymerization of ω -halo α -olefins with

α−olefins)

T 96-10-6, Aluminum, chlorodiethyl-

(catalysts from TiCl3 and ClEt2Al in polymerization, of ω -halo- α -olefins with α -olefins)

IT 97-93-8, Aluminum, triethvl-

(catalysts from TiCl3 and, in polymerization of

ω-halo α-olefins with α-olefins)

IT 1731-04-0, Zinc, bis(3-methylpentyl)-, (S)- 1854-19-9, Zinc, dissobutyl- (catalysts from TiCl4 and, in stereoselective

polymerization of 3,7-dimethyl-1-octene with (S)-3-methyl-1-pentene)

IT 7550-45-0, Titanium chloride, TiCl4

(catalysts from Zn compds. and, in polymerization of 3,7-dimethyl-1-octene with (S)-3-methyl-1-pentene)

II 7647-01-0P, Hydrochloric acid 10035-10-6P, Hydrobromic acid RL: PREP (Preparation)

(formation of, in polymerization of ω -halo α -olefins with α -olefins with Ziegler catalysts)

=> d ibib abs hitstr hitind 132 1-5

L32 ANSWER 1 OF 5 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2008:285253 HCAPLUS Full-text

DOCUMENT NUMBER: 148:309625

TITLE: Arborescent polymers based on at least one

inimer and isoolefin that have been end-functionalized with a polymer or copolymer

having a low glass transition temperature
INVENTOR(S): Kaszas, Gabor; Puskas, Judit; Kulbaba, Kevin;

Nelson, Robert Lloyd
PATENT ASSIGNEE(S): The University of Akron, USA; Lanxess, Inc.

SOURCE: PCT Int. Appl., 76pp.

CODEN: PIXXD2
DOCUMENT TYPE: Patent

DOCUMENT TYPE: LANGUAGE:

English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PAT	PATENT NO.				KIND DATE			1	APPL	ICAT:	ION I	.00		D.	ATE	
WO	 WO 2008027589				A2 20080306			1	WO 2	007-	US19:	280		2	00708	
WO	2008	0275	89		A3		2008	0424							,	1
	W:	CA, ES, JP, LY, NZ, SM, ZA,	CH, FI, KE, MA, OM, SV, ZM,	CN, GB, KG, MD, PG, SY, ZW	CO, GD, KM, ME, PH, TJ,	CR, GE, KN, MG, PL, TM,	AU, CU, GH, KP, MK, PT, TN,	CZ, GM, KR, MN, RO, TR,	DE, GT, KZ, MW, RS, TT,	DK, HN, LA, MX, RU, TZ,	DM, HR, LC, MY, SC, UA,	DO, HU, LK, MZ, SD, UG,	DZ, ID, LR, NA, SE, US,	EC, IL, LS, NG, SG, UZ,	EE, IN, LT, NI, SK, VC,	EG, IS, LU, NO, SL, VN,
PRIORITY		IE, TR, TD, ZM,	IS, BF, TG, ZW,	IT, BJ, BW, AM,	LT, CF, GH,	LU, CG, GM,	CZ, LV, CI, KE, KG,	MC, CM, LS,	MT, GA, MW, MD,	NL, GN, MZ,	PL, GQ, NA, TJ,	PT, GW, SD, TM,	RO, ML, SL, AP,	SE, MR, SZ, EA,	SI, NE, TZ, EP,	SK, SN, UG,

200609

Disclosed is an arborescent polymer comprising: an elastomeric polymer portion having two or more branching points, the arborescent elastomeric polymer block having a low glass transition temperature (Tg), and one or more endfunctionalized portions, wherein one or more end-functionalized portions terminate at least one or two or more branches of the arborescent elastomeric polymer portion of the end-functionalized arborescent polymer. In one embodiment, the present invention relates to arborescent polymers formed from at least one inimer and at least one isoolefin that have been endfunctionalized with a polymer or copolymer having a low glass transition temperature (Tg), and to a process for making such arborescent polymers. In another embodiment, the present invention relates to arborescent polymers formed from at least one inimer and at least one isoolefin that have been endfunctionalized with less than about 5 weight% end blocks derived from a polymer or copolymer having a high glass transition temperature (Tg), and to a process for making such arborescent polymers. In an example of the invention 0.35 g of p-methoxycumylstyrene, 900 cm3 hexane, 600 cm3 Me chloride, 2 cm3 2,6-di-tert-butylpyridine and 240 cm3 isobutylene were added to a reactor, polymerization was started at -95° by addition of 6 cm3 of TiCl4, after 2 h of polymerization a mixture of 236 cm3 isoprene and 150 cm3 Me chloride, 150 cm3 hexane and a small amount of 2,6-di-tert-butylpyridine were added, and the polymerization was continued for another 30 min, before terminating it by adding methanol and NaOH to the mixture, after Me chloride evaporation, washing and isolation of the polymer, the dried weight of the polymer was 164.7 q. Polymers obtained in a similar way were mixed with carbon black, and optionally with a vulcanization agent, and tested as such and with the filler in comparison with regular butyl rubber.

IT 69078-25-7DF, arborescent polymers/synthetic rubbers
477350-05-3DP, arborescent polymers/synthetic rubbers
477350-08-6DP, arborescent polymers/synthetic rubbers
477350-12-2DP, arborescent polymers/synthetic rubbers
RI: IMF (Industrial manufacture); PEP (Physical, engineering or
chemical process); TEM (Technical or engineered material use); PREP
(Preparation); PROC (Process); USES (Uses)
(arborescent polymers based on branched functionalized styrene
inimer and isoolefins, like isobutylene)
RN 69078-25-7 HCAPLUS

1-Hexene, 5-chloro-3,3,5-trimethyl- (CA INDEX NAME)

CN

AB

RN 477350-05-3 HCAPLUS
CN 1-Heptene, 6-chloro-4,4,6-trimethyl- (CA INDEX NAME)

RN 477350-08-6 HCAPLUS

CN 1-Nonene, 8-chloro-4,4,6,6,8-pentamethyl- (CA INDEX NAME)

March 20, 2009 10/559,779 36

RN 477350-12-2 HCAPLUS

CN 1-Octene, 7-chloro-3,3,5,5,7-pentamethyl- (CA INDEX NAME)

CC 39-4 (Synthetic Elastomers and Natural Rubber)

IT Polymerization catalysts

(Lewis acid, halides; arborescent polymers

based on branched functionalized styrene inimer and isoolefins, like isobutylene)

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78-79-5DP, arborescent polymers/synthetic rubbers 100-42-5DP,
arborescent polymers/synthetic rubbers 100-42-5DP, derivs.,
arborescent polymers/synthetic rubbers 106-99-0DP, 1,3-Butadiene,
arborescent polymers/synthetic rubbers 563-46-2DP, arborescent
polymers/synthetic rubbers 592-46-1DP, 2,4-Hexadiene, arborescent polymers/synthetic rubbers 763-29-1DP, arborescent polymers/synthetic rubbers 763-30-4DP, arborescent
polymers/synthetic rubbers 764-13-6DP, arborescent
polymers/synthetic rubbers 1118-58-7DP, arborescent
polymers/synthetic rubbers 2622-21-1DP, arborescent
polymers/synthetic rubbers 3049-88-5DP, arborescent
polymers/synthetic rubbers 4049-81-4DP, arborescent
polymers/synthetic rubbers 4549-74-0DP, arborescent
polymers/synthetic rubbers 5732-01-4DP, 1,3-Butadiene,
2-neopentyl-, arborescent polymers/synthetic rubbers 13643-06-6DP,
arborescent polymers/synthetic rubbers 25924-78-1DP, Piperyline,
arborescent polymers/synthetic rubbers 26519-91-5DP,
Methylcyclopentadiene, arborescent polymers/synthetic rubbers
29797-09-9DP, Cyclohexadiene, arborescent polymers/synthetic rubbers
57908-07-3DP, arborescent polymers/synthetic rubbers
69078-25-7DP, arborescent polymers/synthetic rubbers
140220-46-8DP, arborescent polymers/synthetic rubbers
140220-95-7DP, arborescent polymers/synthetic rubbers
173932-32-6DP, arborescent polymers/synthetic rubbers
477350-02-0DP, arborescent polymers/synthetic rubbers
477350-03-1DP, arborescent polymers/synthetic rubbers
477350-04-2DP, arborescent polymers/synthetic rubbers
477350-05-3DP, arborescent polymers/synthetic rubbers
477350-06-4DP, arborescent polymers/synthetic rubbers
477350-07-5DP, arborescent polymers/synthetic rubbers
477350-08-6DP, arborescent polymers/synthetic rubbers
477350-09-7DP, arborescent polymers/synthetic rubbers
477350-10-0DP, arborescent polymers/synthetic rubbers
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477350-11-1DP, arborescent polymers/synthetic rubbers

477350-12-2DP, arborescent polymers/synthetic rubbers
477350-13-3DP, arborescent polymers/synthetic rubbers
RL: IMF (Industrial manufacture); PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (USEs)

(arborescent polymers based on branched functionalized styrene inimer and isoolefins, like isobutylene)

L32 ANSWER 2 OF 5 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2003:825944 HCAPLUS Full-text

DOCUMENT NUMBER: 140:78789

TITLE: Saucy-Marbet ketonization in a continuous

fixed-bed catalytic reactor

AUTHOR(S): Frauchiger, Simon; Baiker, Alfons

CORPORATE SOURCE: Institute for Chemical and Bioengineering, Swiss

Federal Institute of Technology, ETH Hoenggerberg, Zurich, CH-8093, Switz.

SOURCE: Applied Catalysis, A: General (2003), 253(1), 33-48

CODEN: ACAGE4; ISSN: 0926-860X

PUBLISHER: Elsevier Science B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 140:78789
AB Saucy-Marbet ketonizations of two unsa

Saucy-Marbet ketonizations of two unsatd, alcs., 3.7-dimethyl-oct-6-en-1-vn-3ol (dehydrolinalool (DLL)) and 2-methyl-3-butyn-2-ol (MB) with 2methoxypropene (MP), were studied in a continuous fixed-bed reactor containing a com. alkyl sulfonic acid polysiloxane catalyst, Deloxan ASP I/9, Degussa, a condensate of propyl-3-sulfonic acid-siloxane and SiO2. The solid acid catalyst showed an initial transient period of several hours until stable operation was reached. Similar products were observed as in corresponding homogeneous catalytic reactions indicating similar reaction pathways. However, the heterogeneous catalytic reactions showed lower selectivity to the desired products than the corresponding homogeneous reactions reported in the literature. This outcome is traced to the complex transformation the unsatd. ether MP undergoes upon interaction with the surface of the solid acid catalyst. Furthermore, both Bronsted and Lewis-acid sites were present on the catalyst surface. These factors are the main reasons for the less efficient use of reactants observed in the heterogeneous catalytic reactions. A maximum yield of 54% to the desired product could be achieved in the heterogeneous catalytic ketonization of 3,7-dimethyl-oct-6-en-1-vn-3-ol and 45% for the 2methyl-3-butyn-2-ol ketonization, resp. The studies indicate that the continuous ketonization in a fixed-bed reactor is feasible, but for tech. application, more selective acid catalysts are necessary to compete with the generally used homogeneous catalytic routes.

IT 191930-68-4P

RL: BYP (Byproduct); PREP (Preparation)

(conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet ketonizations of unsatd. alcs.

in continuous fixed-bed reactor)

RN 191930-68-4 HCAPLUS

CN 1-Pentene, 2,4-dimethoxy-4-methyl- (CA INDEX NAME)

- 45-4 (Industrial Organic Chemicals, Leather, Fats, and Waxes) Section cross-reference(s): 67
- ST unsatd alc ketonization methoxypropene heterogeneous catalytic route; dimethyloctenynol ketonization alkylsulfonic acid polysiloxane silica catalyst
- Carbonylation catalysts

(Saucy-Marbet, heterogeneous; conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet ketonizations of unsatd. alcs. in continuous

fixed-bed reactor)

Carbonvlation

(Saucy-Marbet; conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet

ketonizations of unsatd. alcs. in continuous fixed-bed reactor) Lewis acidity

(conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet ketonizations of unsatd. alcs.

in continuous fixed-bed reactor)

Reactors

(fixed-bed; conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet

- ketonizations of unsatd, alcs, in continuous fixed-bed reactor)
- Polysiloxanes, uses
 - RL: CAT (Catalyst use); USES (Uses)

(propyl-3-sulfonic acid containing; conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet ketonizations of unsatd. alcs. in continuous

fixed-bed reactor) 77-76-9P, 2,2-Dimethoxypropane 141-10-6P 53654-62-9P

191930-68-4P RL: BYP (Byproduct); PREP (Preparation)

(conversion and selectivity of sulfonic acid polysiloxane/silica

catalyst in Saucy-Marbet ketonizations of unsatd. alcs. in continuous fixed-bed reactor)

7631-86-9, Silica, uses 640735-92-8, Deloxan ASP-I 9 RL: CAT (Catalyst use); USES (Uses)

(conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet ketonizations of unsatd. alcs. in continuous fixed-bed reactor)

115-19-5, 2-Methvl-3-butvn-2-ol 116-11-0 29171-20-8, 3,7-Dimethyl-oct-6-en-1-yn-3-ol

RL: EPR (Engineering process); PEP (Physical, engineering or chemical process); RCT (Reactant); PROC (Process); RACT (Reactant or reagent)

(conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet ketonizations of unsatd. alcs.

in continuous fixed-bed reactor)

1604-28-0P, 6-Methyl-3,5-heptadien-2-one 13837-60-0P, 2-Ethynyl-tetrahydro-2,6,6-trimethyl-2H-pyran 16647-01-1P, 6-Methyl-4,5-heptadien-2-one 16647-05-5P.

6,10-Dimethyl-4,5,9-undecatrien-2-one

RL: IMF (Industrial manufacture); PREP (Preparation)

(conversion and selectivity of sulfonic acid polysiloxane/silica catalyst in Saucy-Marbet ketonizations of unsatd. alcs. in continuous fixed-bed reactor)

REFERENCE COUNT:

THERE ARE 29 CITED REFERENCES AVAILABLE 29 FOR THIS RECORD, ALL CITATIONS AVAILABLE IN THE RE FORMAT

L32 ANSWER 3 OF 5 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1999:719035 HCAPLUS Full-text

English

DOCUMENT NUMBER: 131:337525

TITLE: Silyl-functional initiator for living cationic

polymerization

INVENTOR(S): Faust, Rudolf; Hadjikyriacou, Savvas E.; Suzuki,

PATENT ASSIGNEE(S):

University of Massachusetts, USA; Dow Corning

Corporation SOURCE: U.S., 7 pp.

CODEN: USXXAM Patent

DOCUMENT TYPE: LANGUAGE:

FAMILY ACC. NUM. COUNT: 1

DATENT INFORMATION.

PAIDNI INFOR	MATION:				
PATENT	NO.	KIND	DATE	APPLICATION NO.	DATE
US 5981	- 785	A	19991109	US 1998-204900	199812
WO 2000	032609	A1	20000608	WO 1999-US27461	03 199911 19
	CZ, DE, DK, IN, IS, JP, MD, MG, MK, SI, SK, SL, AZ, BY, KG, GH, GM, KE,	EE, ES, KE, KG, MN, MW, TJ, TM, KZ, MD, LS, MW,	FI, GB, KP, KR, MX, NO, TR, TT, RU, TJ, SD, SL,	BB, BG, BR, BY, CA, GD, GE, GH, GM, HR, KZ, LC, LK, LR, LS, NZ, PL, PT, RO, RU, UA, UG, UZ, VN, YU, TM SZ, TZ, UG, ZW, AT, IE, IT, LU, MC, NL,	CH, CN, CU, HU, ID, IL, LT, LU, LV, SD, SE, SG, ZA, ZW, AM, BE, CH, CY,
PRIORITY APP	BJ, CF, CG,			GW, ML, MR, NE, SN, US 1998-204900	TD, TG

OTHER SOURCE(S): MARPAT 131:337525

AB Haloorganosilane initiators which can be reacted with cationically polymerizable monomers in the presence of a Lewis acid to provide living polymers having a silv1-functional head group have the structure XaR3aSiZCR2R1, where X is halogen, R is independently selected from alkyl groups having 1-10 carbon atoms or aryl groups having 6-10 carbon atoms, Z is a divalent aliphatic hydrocarbon group having at least 3 carbon atoms, R1 is halogen, alkoxy, acyloxy or hydroxy, and a is 1, 2, or 3. Thus, 3,3,5trimethyl-5-chloro-1-hexene was treated with dichloromethylsilane in the presence of a platinum-divinyltetramethyldisiloxane complex catalyst to give a hydrosilated product which was rehydrochlorinated by treatment with HCl to give 1-methyldichlorosily1-3,3,5-trimethyl-5- chlorohexane (I). Isobutylene was polymerized in a MeCl/hexane solvent system using I, titanium tetrachloride, and 2,6-di-tert-butylpyridine as catalysts. The resulting cationic living polymer having number-average mol. weight 3200 was coupled using 2,5-bis(2-furylpropyl)furan as the coupling agent to give a polymer which on treatment with MeOH gave a methoxysilyl-functional telechelic polyisobutylene (II) having number-average mol. weight 5300. II was cured by exposure to moisture in the presence of tin octoate and could be used in the formulation of sealants, adhesives and coatings.

TT 69078-25-7 (reactant; in preparation of haloorganosilane initiators for living cationic polymerization of vinyl monomers)

69078-25-7 HCAPLUS RN

CN 1-Hexene, 5-chloro-3,3,5-trimethyl- (CA INDEX NAME)

ICM C07F007-12

INCL 556488000

CC 35-3 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 39

haloorganosilane initiator living cationic polymn; isobutylene living cationic polymn silane catalyst; moisture curable polvisobutylene

Polymerization

Polymerization catalysts

(cationic, living; reactive silyl-functional initiators for living cationic polymerization of vinyl monomers and preparation of polymers

containing hydrolyzable silyl end groups)

Hydrosilylation catalysts

(platinum organosiloxane complex; for preparation of silyl-functional initiators for living cationic polymerization of vinyl compds.)

7440-06-4D, Platinum, complex with divinyltetramethyldisiloxane,

30110-75-9D, Divinyltetramethyldisiloxane, platinum complex RL: CAT (Catalyst use); USES (Uses)

(hydrosilylation catalyst; for preparation of

silyl-functional initiators for living cationic polymerization of vinyl compds.)

249905-14-4P

RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)

(polymerization catalyst; in polymerization of isobutylene)

TТ 249905-12-2P

RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP

(Preparation); USES (Uses)

(polymerization catalyst; reactive silyl-functional initiators for living cationic polymerization of vinyl monomers and preparation of polymers containing hydrolyzable silyl end groups)

75-54-7, Dichloromethylsilane 78-79-5, Isoprene, reactions 69078-25-7

RL: RCT (Reactant): RACT (Reactant or reagent)

L32 ANSWER 4 OF 5 HCAPLUS COPYRIGHT 2009 ACS on STN

(reactant; in preparation of haloorganosilane initiators for living cationic polymerization of vinvl monomers)

REFERENCE COUNT:

14 THERE ARE 14 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

ACCESSION NUMBER: 1989:155017 HCAPLUS Full-text DOCUMENT NUMBER: 110:155017

ORIGINAL REFERENCE NO.: 110:25659a,25662a

TITLE: Initiators and process for living polymerization of olefins to end-functionalized polymers

INVENTOR(S): Kennedy, Joseph Paul; Mishra, Munmaya Kumar

PATENT ASSIGNEE(S): SOURCE:

Eur. Pat. Appl., 34 pp. CODEN: EPXXDW

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PA	TENT NO.	KIND	DATE	Al	PPLICATION NO.	DATE
EP	265053	A2	19880427	E	P 1987-307508	
						198708 25
ED	265053	3.2	19900829			23
			19961023			
EP						
	R: AT, BE, CH,					
CA	1338520	С	19960813	C	A 1987-545335	
						198708 25
AT	144542	Т	19961115	Δ.	т 1987-307508	
		-				198708
						25
.TD	01062308	A	19890308	.11	P 1987-216542	20
0.2	01002300	••	13030300	٠.	1 1907 210012	198709
						01
TD	2707086	B2	19980128			01
	4929683	A A	19900120		S 1988-173049	
US	4929683	A	19900529	U,	5 1988-1/3049	198803
		_				28
US	5066730	A	19911119	U	S 1990-500771	
						199003
						28
PRIORIT	Y APPLN. INFO.:			U	S 1986-899655 A	
						198608
						25
				U	S 1988-173049 A3	3
						198803
						28

- AB The title living vinvl polymers, having specific end groups and a wide range of mol. wts., are prepared by cationic living polymerization of vinyl monomers [e.g., isobutylene (I) or styrene] in the presence of complexes of organic ethers and Lewis acids comprising BCl3, AlCl3, SnCl4, TiCl4, SbF5, FeCl3, ZnCl2, and VCl4. Cumyl Me ether (II) 7.7 + 10-2, I 1.17, and BCl3 0.37 mol/L were stirred in 20 mL CH2C12 at -30° for 0.5 h to give a I polymer terminated with C6H5C(CH3)2 and C1 and having number-average mol. weight (Mn) 1500, weight-average mol. weight Mw 5000, and Mn/Mw 3.3. A polymer having Mn 13,700; Mw 30,000, and Mn/Mw 2.2 was obtained when 7.4 + 10-3 mol/L II was used.
- 119886-65-6
- RL: USES (Uses)
- (initiator, for preparation of cationic living polymer systems) 119886-65-6 HCAPLUS RN
- CN Boron, trichloro(4-methoxy-1-pentene)-, (T-4)- (CA INDEX NAME)

March 20, 2009 10/559,779 42

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Me-CH-CH2-CH2-CH2
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IC ICM C08F010-00 ICS C08F004-06

CC 35-3 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 67 ether trichloroboron complex polyisobutylene initiator; boron trichloro ether complex initiator; cationic polymn Lewis

acid ether complex; chlorine terminated polyisobutylene; catalyst living polymn isobutylene; styrene living polymn catalyst; mol wt control polyisobutylene catalyst;

polydispersity polyisobutylene ether Lewis acid; chain transfer agent polyisobutylene

IT Polymerization catalysts

(ether-Lewis acid compds., for preparation of living cationic polymer systems)

Polymerization

(cationic, living, of olefins, ether-Lewis acid compds. for)

Ethers, compounds

RL: USES (Uses)

(compds., with Lewis acids, initiators, for living cationic polymerization of olefins)

Lewis acids

RL: USES (Uses)

DOCUMENT NUMBER:

(compds., with ethers, initiators, for living cationic polymerization of olefins)

Polymers, preparation

RL: IMF (Industrial manufacture); PREP (Preparation) (telechelic, manufacture of, having low polydispersity, Lewis acid-ether compound initiators for)

58286-97-8 119865-82-6 119865-83-7 119865-84-8 119886-47-4 119886-48-5 119886-49-6 119886-50-9 119886-51-0 119886-52-1 119886-53-2 119886-54-3 119886-55-4 119886-56-5 119886-57-6 119886-58-7 119886-59-8 119886-60-1 119886-61-2 119886-62-3 119886-63-4 119886-64-5 119886-65-6 119886-66-7 119886-67-8 119886-68-9 119886-69-0 119886-70-3 119886-71-4 119886-72-5 RL: USES (Uses)

(initiator, for preparation of cationic living polymer systems)

9010-85-9P RL: IMF (Industrial manufacture); PREP (Preparation)

(manufacture of, with low polydispersity, Lewis acid -ether compound polymerization initiators for)

9003-27-4DP, Polyisobutylene, chlorine and tertiary hydrocarbyl-terminated 9003-53-6DP, Polystyrene, chlorine and tertiary hydrocarbyl-terminated

RL: IMF (Industrial manufacture); PREP (Preparation) (manufacture of, with low polydispersity, Lewis acid -ether complex polymerization initiators for)

L32 ANSWER 5 OF 5 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1965:446640 HCAPLUS Full-text 63:46640

ORIGINAL REFERENCE NO.: 63:8497h,8498a-b

TITLE: Copolymerization of α -olefins with

 ω -halo α -olefins by use of Ziegler

author(S): catalysts
Bacskai, R.

CORPORATE SOURCE: California Res. Corp., Richmond

SOURCE: Journal of Polymer Science, Part A: General

Papers (1965), 3(7), 2491-510 CODEN: JPYAAK; ISSN: 0449-2951

DOCUMENT TYPE: Journal

LANGUAGE: English

AB The copolymerization of α -olefins with 5-chloro-1-pentene, 6-chloro-1-hexene, 8-chloro-1-octene, 7-chloro-1-octene, 3-bromo-1-octene, 4-bromo-1-butene, 5bromo-1-pentene, 6-bromo-1-hexene, and 11-chloro-1-undecene by use of Ziegler cstalvsts has been investigated. ω -Halo α -olefins deactivate the catalvst, thus resulting in low copolymer yield and reduced mol. weight The deactivation is caused by H halide formation which is greatly facilitated in the case of the allyl halides and 6-halo-1-hexenes. The different degrees of deactivation were interpreted by considering the ease of carbonium ion formation from the various ω -halo α -olefins. The copolymer yields were considerably improved when a catalytic amount of a pyridine, Et3N, or tetrahydrofuran was added to the polymerization system. It is proposed that the added Lewis base, by decreasing the Lewis acid character of the catalyst, suppresses the decomposition of the ω -halo α -olefin leading to H halide formation. Depending on the Ziegler catalyst employed, the copolymers of α olefins and ω -halo α -olefins can be either crystalline or amorphous solids. The halogen, especially the Br in the copolymers, is reactive in nucleophilic substitution reactions. Reaction with amines gave polymers which could be dyed by acid dyes.

T 31950-56-8

(Derived from data in the 7th Collective Formula Index (1962-1966))

RN 31950-56-8 HCAPLUS

CN 1-Pentene, 4-bromo- (CA INDEX NAME)

Me-CH-CH2-CH-CH2

CC 45 (Synthetic High Polymers)

IT Bases

(Lewis, catalysts from Al compds., TiCl3 and, in polymerization of $\omega-$ halo $\alpha-$ olefins with

α-olefins) Catalysts and Catalysis

(in polymerization, of ω -halo α -olefins with α -olefins, Ziegler)

IT Polymerization

(of $\omega\text{-halo}$ $\alpha\text{-olefins}$ with $\alpha\text{-olefins},$ Ziegler catalysts in)

IT Olefins

(polymerization of α -, with ω -halo α -olefins, Ziegler catalysts in)

IT Olefins

(ω-halo α-, polymerization with α-olefins,

Ziegler catalysts in)

T 7705-07-9, Titanium chloride, TiCl3 (catalysts from Al compds. and TiCl3 in polymerization, of m-halo a-olefins with a-olefins)

IT 109-99-9, Furan, tetrahydro- 110-86-1, Pyridine 121-44-8, Triethylamine

(catalysts from Al compds., TiCl3 and, in polymerization of ω -halo α -olefins with α -olefins)

IT 96-10-6, Aluminum, chlorodiethyl-(catalysts from TiCl3 and ClEt2Al in polymerization, of o-halo-a-olefins with a-olefins)

IT 97-93-8, Aluminum, triethyl-(catalysts from Ticl3 and, in polymerization of φ-halo α-olefins with α-olefins)

IT 1731-04-0, Zinc, bis(3-methylpentyl)-, (S)- 1854-19-9, Zinc,
 disobutyl (catalysts from TiCl4 and, in stereoselective

polymerization of 3,7-dimethyl-1-octene with (S)-3-methyl-1-pentene)

IT 7550-45-0, Titanium chloride, TiCl4

(catalysts from Zn compds. and, in polymerization of 3,7-dimethyl-1-octene with (S)-3-methyl-1-pentene)

IT 7647-01-0P, Hydrochloric acid 10035-10-6P, Hydrobromic acid RL: PREP (Preparation)

(formation of, in polymerization of $\omega-\text{halo}$ $\alpha-\text{olefins}$ with $\alpha-\text{olefins}$ with Ziegler catalysts)